



Name: _____

Date: _____

Student Exploration: Limiting Reactants

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

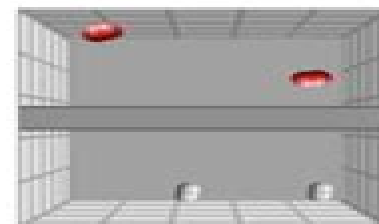
Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make? _____ you can make 30 hotdogs with single bun and wieners _____
2. Which ingredient limited the number of hot dogs you could make? the buns having 30 limited the number of hot dog i can make _____
3. Which ingredient will you have leftovers of? i would have leftover of the wieners remaining 2 more. _____

Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.



To begin, make sure $\text{H}_2 + \text{O}_2$ becomes H_2O . The small "2" in H_2 , O_2 , and H_2O is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

1. Use the sliders to set the number of **O_2 molecules** and **H_2 molecules** to two.
 - A. How many hydrogen molecules (H_2) are there? _____ 1 _____ Hydrogen atoms? _____ 2 _____
 - B. How many oxygen molecules (O_2) are there? _____ 1 _____ Oxygen atoms? _____ 2 _____
2. How many H_2O molecules do you think will form when these four molecules react? _____ 4 _____



Answers For Student Limiting Reactants

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Table of Contents Answers For Student Limiting Reactants

1. Understanding the eBook Answers For Student Limiting Reactants
 - The Rise of Digital Reading Answers For Student Limiting Reactants
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Student Limiting Reactants
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Student Limiting Reactants
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Student Limiting Reactants
 - Personalized Recommendations
 - Answers For Student Limiting Reactants User Reviews and Ratings

- Answers For Student Limiting Reactants and Bestseller Lists
- 5. Accessing Answers For Student Limiting Reactants Free and Paid eBooks
 - Answers For Student Limiting Reactants Public Domain eBooks
 - Answers For Student Limiting Reactants eBook Subscription Services
 - Answers For Student Limiting Reactants Budget-Friendly Options
- 6. Navigating Answers For Student Limiting Reactants eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Student Limiting Reactants Compatibility with Devices
 - Answers For Student Limiting Reactants Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Student Limiting Reactants
 - Highlighting and Note-Taking Answers For Student Limiting Reactants
 - Interactive Elements Answers For Student Limiting Reactants
- 8. Staying Engaged with Answers For Student Limiting Reactants
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Student Limiting Reactants
- 9. Balancing eBooks and Physical Books Answers For Student Limiting Reactants
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Student Limiting Reactants
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Student Limiting Reactants
 - Setting Reading Goals Answers For Student Limiting Reactants
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Student Limiting Reactants
 - Fact-Checking eBook Content of Answers For Student Limiting Reactants
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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