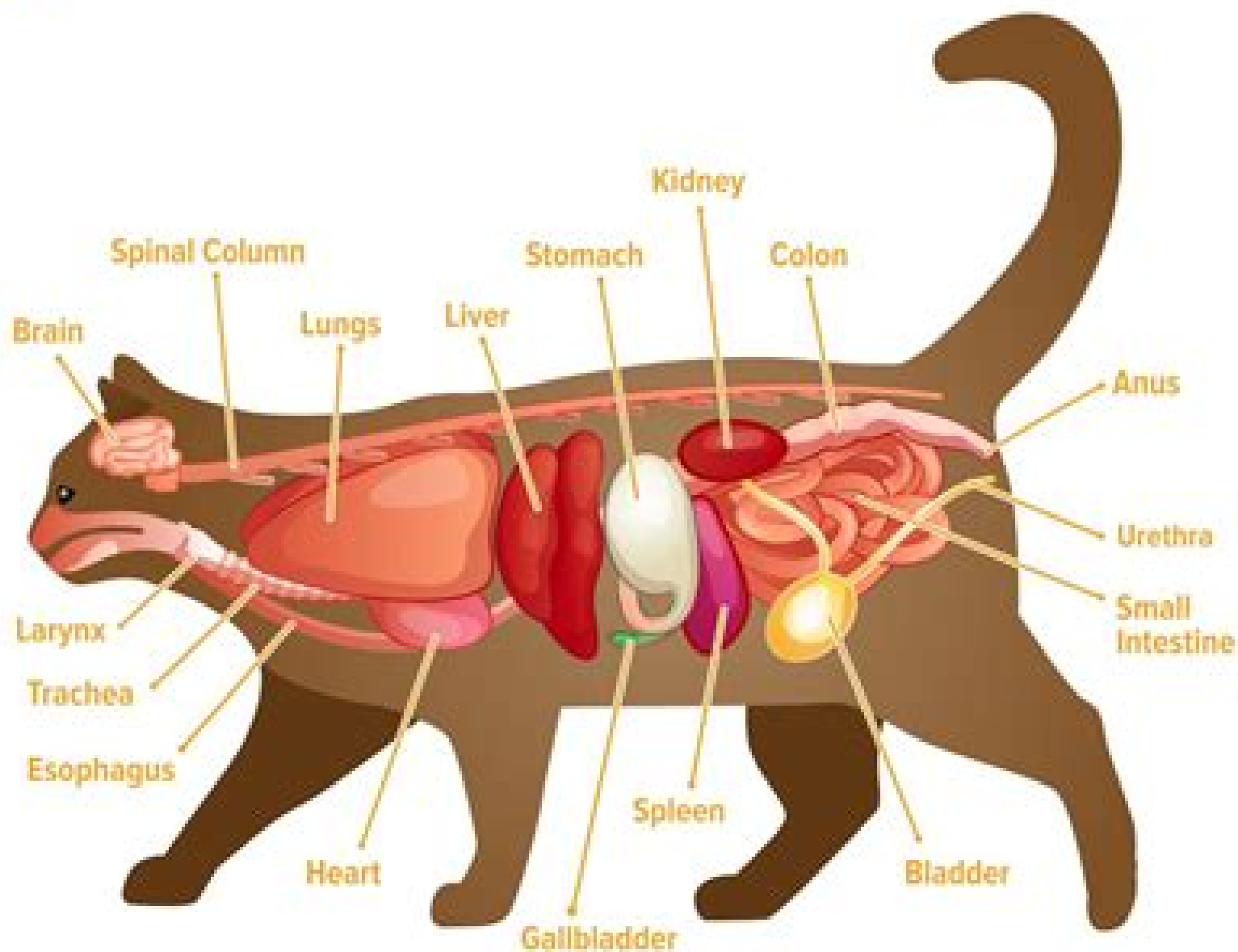




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web what factors determine the explosiveness of the reaction of hydrogen with oxygen in this lab we will generate microscale quantities of hydrogen and oxygen and test their explosive nature first separately then in mixtures of various proportions the goal to find the most powerful gas mixture and use it to launch a rocket across the room

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web other descriptions of the lab can be easily found if you do a web search for micro rocket lab or mole rocket lab there is a time investment to build the nozzles for the gas generators and the piezo ignition devices but after that initial investment lab set up is fast and simple in future years i have attached the handout i give

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web in this lesson i work through the flinn mini mole rocket lab in this lab you will investigate limiting and excess reactants there is no need for a write up

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web this lab is an excellent an introduction to limiting excess reagents where students turn the bulb of a plastic beaker style

pipette into a micro scale rocket different ratios of hydrogen oxygen are collected in the bulb and the most explosive mixture used to shoot the rocket across the room

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web may 22 2019 dustin williams wed 05 22 2019 20 13 there is a lab that is called something like the mole rocket lab or micro rockets some of you may be familiar with the lab but i wanted to write this post to share it with teachers who may not be aware of it

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web stoichiometry limiting reactants background hydrogen the most abundant element in the universe is a colorless odorless gas it is combustible which means that it burns quite readily hydrogen gas is conveniently generated in the lab by the reaction of zinc metal with hydrochloric acid

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web step 7 launch the rocket once the desired amount of gas is put into the rocket carefully put the rocket over top of the ignition apparatus make sure that there is still a small amount of water in the bottom to act as a propellant make sure the way is clear do a count down and spark the rocket

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