

Quiz & Worksheet - Isotopes and Average Atomic Mass

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1. How are isotopes of the same element similar?

- ☐ They have the same particles in the nucleus.
- ☐ They are equally abundant in nature.
- ☐ They have the same mass.
- ☐ They have the same number of neutrons.
- ☐ They have the same number of protons.

2. How are isotopes of the same element different?

- ☐ They will have undergo different chemical reactions.
- ☐ They have different numbers of protons.
- ☐ They have different numbers of neutrons.
- ☐ They will have different locations on the periodic table.
- ☐ They will have different numbers of electrons.

3. How many neutrons does the most common isotope of hydrogen have?

- ☐ 3
- ☐ 2
- ☐ 1
- ☐ 0
- ☐ 4

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Yijin Wang



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