

Sodium becomes a cation, and chlorine becomes an anion.

Activity A: Ions	Get the Gizmo ready: - Click Reset. - Check that sodium and chlorine are still selected.	
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Introduction: Some of the particles that make up atoms have an electrical charge. Electrons are negatively charged, while protons are positively charged. Particles with opposite charges (+ and -) attract, while particles with the same charge (+ and + or - and -) repel.

Question: What happens when atoms gain or lose electrons?

1. **Count:** Electrons move around the nucleus of atoms in specific shells, shown by the rings around the atoms in the Gizmo. The first ring holds two electrons, and the second holds eight. (Electrons in the inner rings are not shown; you can assume these rings are full.)

- A. Observe the sodium and chlorine atoms. Assuming that the inner rings are full of electrons, how many electrons are there total in each atom?

Sodium: 11 electrons

Chlorine: 17 electrons

- B. Each atom is neutrally charged, which means that each atom has the same number of protons and electrons. Based on this, how many protons are in each atom?

Sodium: 11 protons

Chlorine: 17 protons

2. **Observe:** Most atoms are stable with a configuration of eight valence electrons. This is known as the **octet rule**. How many valence electrons does each atom have?

Sodium: 1 valence electron

Chlorine: 7 valence electrons

3. **Form a bond:** Each electron has a charge of $1-$, and each proton has a charge of $1+$. You can calculate the charge of an atom by subtracting the number of electrons from the number of protons. Move an electron from the sodium to the chlorine atom.

- A. What are the charges of each atom now? Sodium: +1 Chlorine: -1

Turn on **Show charge** to check. These charged atoms are called **ions**.

- B. Is each ion stable? Explain.

Yes, each ion is stable. Chlorine only has 7 valence electrons, so gaining 1 more would lead to a stable full octet. Sodium on the other hand has 1 valence electron, by losing this electron it has just a regular shell of 8 electrons, also becoming a stable octet.

Click **Check** in the lower right corner of the Gizmo to check.

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10th Ionic Bonds Gizmo Answers

S Marginson

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

10th Ionic Bonds Gizmo Answers:

Reviewing **10th Ionic Bonds Gizmo Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**10th Ionic Bonds Gizmo Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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Table of Contents **10th Ionic Bonds Gizmo Answers**

1. Understanding the eBook 10th Ionic Bonds Gizmo Answers
 - The Rise of Digital Reading 10th Ionic Bonds Gizmo Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying 10th Ionic Bonds Gizmo Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 10th Ionic Bonds Gizmo Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 10th Ionic Bonds Gizmo Answers
 - Personalized Recommendations
 - 10th Ionic Bonds Gizmo Answers User Reviews and Ratings
 - 10th Ionic Bonds Gizmo Answers and Bestseller Lists

5. Accessing 10th Ionic Bonds Gizmo Answers Free and Paid eBooks
 - 10th Ionic Bonds Gizmo Answers Public Domain eBooks
 - 10th Ionic Bonds Gizmo Answers eBook Subscription Services
 - 10th Ionic Bonds Gizmo Answers Budget-Friendly Options
6. Navigating 10th Ionic Bonds Gizmo Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - 10th Ionic Bonds Gizmo Answers Compatibility with Devices
 - 10th Ionic Bonds Gizmo Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 10th Ionic Bonds Gizmo Answers
 - Highlighting and Note-Taking 10th Ionic Bonds Gizmo Answers
 - Interactive Elements 10th Ionic Bonds Gizmo Answers
8. Staying Engaged with 10th Ionic Bonds Gizmo Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 10th Ionic Bonds Gizmo Answers
9. Balancing eBooks and Physical Books 10th Ionic Bonds Gizmo Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 10th Ionic Bonds Gizmo Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 10th Ionic Bonds Gizmo Answers
 - Setting Reading Goals 10th Ionic Bonds Gizmo Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 10th Ionic Bonds Gizmo Answers
 - Fact-Checking eBook Content of 10th Ionic Bonds Gizmo Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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