

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

1. Explain how a chemical equation is written.
2. Explain how to write a balanced equation.
3. Write the steps for writing a balanced chemical equation.

Vocabulary

- | | |
|----------------------|----------------------|
| 1. chemical equation | 4. coefficient |
| 2. stoichiometric | 5. balanced equation |
| 3. reactant | |

Part A Completion

Use the vocabulary words in each sentence, writing either the correct word form or the correct chemical equation. Write each answer on a separate sheet of paper, showing your work.

1. A chemical equation that includes information about the _____ of the reactants, the products, and the states of matter is called a _____ equation. The word _____ refers to the relative amounts of the _____ involved in the reaction.
2. A chemical equation is _____ if the number of atoms of each element is the same on both sides of the equation. To write a balanced equation, you must _____ the equation by changing the _____ of the reactants and products.

3. The chemical equation for the reaction of hydrogen and oxygen to form water is _____ (a) _____ (b) _____ and _____ (c) _____ (d) _____ (e) _____ (f) _____ (g) _____ (h) _____ (i) _____ (j) _____ (k) _____ (l) _____ (m) _____ (n) _____ (o) _____ (p) _____ (q) _____ (r) _____ (s) _____ (t) _____ (u) _____ (v) _____ (w) _____ (x) _____ (y) _____ (z) _____

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Answer Key

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