

Name _____ Date _____ Period _____

Stoichiometry Practice Worksheet

- 1) Balance the equation and then answer the following questions:



- a) How many moles of water will be formed if you start with 2.54 moles of sodium hydroxide and you have an excess of sulfuric acid?
- b) How many grams of sodium sulfate will be formed if you start with 5.00 moles of sodium hydroxide and you have an excess of sulfuric acid?
- c) How many grams of sulfuric acid are required to make 101 g of sodium sulfate? (assuming you have an excess of sodium hydroxide.)

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James C. Hill, Bruce Edward Bursten



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