

3 - 4 Review and Reinforcement

Elements and Compounds

Fill in each blank with the name of the element.

Elemental Matter	Elemental Compound	Elemental & Compound
1. Sodium	Na	Sodium
2. Magnesium	Mg	Magnesium
3. Carbon	C	Carbon
4. Oxygen	O	Oxygen
5. Iron	Fe	Iron
6. Zinc	Zn	Zinc
7. Potassium	K	Potassium
8. Copper	Cu	Copper
9. Hydrogen	H	Hydrogen
10. Nitrogen	N	Nitrogen
11. Chlorine	Cl	Chlorine
12. Sulfur	S	Sulfur

Complete the following by inserting the missing words appropriate to each blank line from the list.

element
compound
mixture

substance
mixture
element

physical matter
matter
chemical matter

1. Elements are matter composed of identical atoms.
2. Elements are physically mixed with each other, which cannot be separated.
3. Elements are composed of pure particles.
4. The mixture of elements chemically combined is called a compound.
5. Chemical is a mixture of an element and a compound is a mixture of a compound.
6. Elements and compounds are called pure substances because they have a unique set of chemical and physical properties.

11 3 Review And Reinforcement Answers

SJ Ball



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question answer and explanation to ensure coverage of the major advances in the classification of child and adolescent disorders represented in the DSM IV

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