

HIMACHAL PRADESH BOARD OF SCHOOL EDUCATION, DHARAMSHALA
Model Question Paper
Class – 11th

Duration – 03:00 Hr

Chemistry(Theory)

M.M.: 60

Instructions:

- i) The question paper consists of four sections- A, B, C and D.*
- ii) Internal choices are given in some questions.*
- iii) Section A contains 12 MCQ 1 to 12 of 1 mark each.*
- iv) Section B contains 9 questions from 13 to 21 of 2 marks each.*
- v) Section C contains 6 questions from 22 to 27 of 3 marks each.*
- vi) Section D contains 3 questions from 28 to 30 of 4 marks each.*
- vii) Make neat and clean diagrams where required.*

Section A (1X12)

Q.1 One mole of oxygen molecule is equal to

- a) 6.023×10^{23} atoms of oxygen b) 6.023×10^{23} atoms of oxygen
c) 6.023×10^{23} molecules of oxygen d) 6.023×10^{23} molecules of oxygen

Q.2 3d orbitals have values

- a) $n=2, l=3$ b) $n=3, l=0$
c) $n=3, l=1$ d) $n=3, l=2$

Q. 3 Which of the following elements has maximum electron gain enthalpy

- a) F b) Cl c) Br d) I

Q.4 Pyramidal geometry associated with

- a) CH_4 b) NH_3 c) H_2O d) CO_2

Q.5 Decreasing order of bond angle in H_2O , NH_3 and CH_4

- a) $\text{CH}_4 > \text{NH}_3 > \text{H}_2\text{O}$ b) $\text{NH}_3 > \text{H}_2\text{O} > \text{CH}_4$
c) $\text{CH}_4 > \text{H}_2\text{O} > \text{NH}_3$ d) $\text{H}_2\text{O} > \text{CH}_4 > \text{NH}_3$

Q.6 According to Boyle's law

- a) At constant T, $V \propto P$ b) At constant T, $V \propto 1/P$
c) At constant P, $V \propto T$ d) At constant T, $P \propto T$

Q.7 For the reaction $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightleftharpoons 2\text{NH}_3(\text{g})$

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Priti Singhal



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Table of Contents 11 Th State Board Chemistry Guide

1. Understanding the eBook 11 Th State Board Chemistry Guide
 - The Rise of Digital Reading 11 Th State Board Chemistry Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying 11 Th State Board Chemistry Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 11 Th State Board Chemistry Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from 11 Th State Board Chemistry Guide
 - Personalized Recommendations
 - 11 Th State Board Chemistry Guide User Reviews and Ratings
 - 11 Th State Board Chemistry Guide and Bestseller Lists
5. Accessing 11 Th State Board Chemistry Guide Free and Paid eBooks
 - 11 Th State Board Chemistry Guide Public Domain eBooks
 - 11 Th State Board Chemistry Guide eBook Subscription Services
 - 11 Th State Board Chemistry Guide Budget-Friendly Options
6. Navigating 11 Th State Board Chemistry Guide eBook Formats

- ePub, PDF, MOBI, and More
 - 11 Th State Board Chemistry Guide Compatibility with Devices
 - 11 Th State Board Chemistry Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of 11 Th State Board Chemistry Guide
 - Highlighting and Note-Taking 11 Th State Board Chemistry Guide
 - Interactive Elements 11 Th State Board Chemistry Guide
8. Staying Engaged with 11 Th State Board Chemistry Guide
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 11 Th State Board Chemistry Guide
9. Balancing eBooks and Physical Books 11 Th State Board Chemistry Guide
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection 11 Th State Board Chemistry Guide
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 11 Th State Board Chemistry Guide
- Setting Reading Goals 11 Th State Board Chemistry Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 11 Th State Board Chemistry Guide
- Fact-Checking eBook Content of 11 Th State Board Chemistry Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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