

Q 3. Answer the following questions.

- (i) Discuss law of Reciprocal Proportion with examples.
- (ii) An inorganic salt gave the following percentage composition:
 $\text{Na} = 29.11\%$, $\text{S} = 40.51\%$ & $\text{O} = 30.38\%$, Calculate the Empirical Formula of the salt.
- (iii) Discuss Aufbau's Principle and Hund's Rule.
- (iv) What is diagonal relationship with example.
- (v) Explain sp^2 hybridization with example.
- (vi) On the basis of VSEPR theory, predict the shapes of the following molecules/ions?

(a) SF_4	(b) NH_4^+	(c) H_3O^+
(d) NH_3	(e) PCl_5	(f) NH_2^-
- (vii) State & explain Hess's Law of constant heat summation.
- (viii) What is Buffer Solution. Give its types.
- (ix) Give three methods of preparation of Alkynes.

SECTION D

Q 4. Answer the following questions.

- (i) What are Quantum numbers? Explain
Or
State and Explain Heisenberg's Uncertainty Principle.
- (ii) Define 1st Law of Thermodynamics. Derive its mathematical formulation.
Or
Define Heat Capacity. Derive relation between C_p & C_v .
- (iii) Explain Nucleophilic Substitution Reaction.
Or
Write the IUPAC names of the following organic compounds:

(a) $\text{CH}_3 - \text{CH}_2 - \text{CH} = \text{CH} - \text{CHO}$
(b) $\begin{array}{c} \text{COOH} \\ \\ \text{COOH} \end{array}$

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1. Understanding the eBook 11th Question Paper Chemistry 23
 - The Rise of Digital Reading 11th Question Paper Chemistry 23
 - Advantages of eBooks Over Traditional Books
2. Identifying 11th Question Paper Chemistry 23
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 11th Question Paper Chemistry 23
 - User-Friendly Interface
4. Exploring eBook Recommendations from 11th Question Paper Chemistry 23
 - Personalized Recommendations
 - 11th Question Paper Chemistry 23 User Reviews and Ratings
 - 11th Question Paper Chemistry 23 and Bestseller Lists
5. Accessing 11th Question Paper Chemistry 23 Free and Paid eBooks
 - 11th Question Paper Chemistry 23 Public Domain eBooks
 - 11th Question Paper Chemistry 23 eBook Subscription Services
 - 11th Question Paper Chemistry 23 Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - 11th Question Paper Chemistry 23 Compatibility with Devices
 - 11th Question Paper Chemistry 23 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 11th Question Paper Chemistry 23
 - Highlighting and Note-Taking 11th Question Paper Chemistry 23
 - Interactive Elements 11th Question Paper Chemistry 23
8. Staying Engaged with 11th Question Paper Chemistry 23
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 11th Question Paper Chemistry 23
9. Balancing eBooks and Physical Books 11th Question Paper Chemistry 23
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 11th Question Paper Chemistry 23
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 11th Question Paper Chemistry 23
 - Setting Reading Goals 11th Question Paper Chemistry 23
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 11th Question Paper Chemistry 23
 - Fact-Checking eBook Content of 11th Question Paper Chemistry 23
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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