

12.3 Limiting Reagent and Percent Yield



BY:
MICHELLE ALICE GANIAN

Reaction Stoichiometry

Amount of Product Formed and Percent Yield



Theoretical Yield:	0.074 g O ₂	
Actual Yield:	0.069 g O ₂	
Percent Yield:	$\frac{0.069 \text{ g O}_2}{0.074 \text{ g O}_2}$	= 93.2% Yield

12 3 Limiting Reagent And Percent Yield Workbook Answers

Alexander Stone Macnow



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