

- (ix) The standard potential of the cell in the following reaction is _____
 $\text{Cd}_{(s)} + \text{Cu}_{(aq)}^{2+} \longrightarrow \text{Cd}_{(aq)}^{2+} + \text{Cu}_{(s)}$
 ($E_{\text{Cd}^{2+}}^{\circ} = -0.403 \text{ V}$, $E_{\text{Cu}^{2+}}^{\circ} = 0.334 \text{ V}$)
 (a) -0.737 V (b) 0.737 V
 (c) -0.069 V (d) 0.069 V
- (x) The value of $[\text{H}_3\text{O}^+]$ in mol lit^{-1} of 0.001 M acetic acid solution ($K_a = 1.8 \times 10^{-5}$) is _____
 (a) 1.34×10^{-1} (b) 1.34×10^{-2}
 (c) 1.34×10^{-3} (d) 1.34×10^{-4}

Q.2. Answer the following questions:

[8]

- Write the product formed when alkyl halide reacts with silver nitrite.
- Write the name of product formed, when acetone is treated with 2, 4-dinitrophenyl hydrazine.
- Write the name of biodegradable polyamide copolymer.
- Identify the molecularity of following elementary reaction:
 $\text{NO}_{(g)} + \text{O}_{3(g)} \longrightarrow \text{NO}_{2(g)} + \text{O}_{2(g)}$
- What is the action of selenium on magnesium metal?
- Write the name of isomerism in the following complexes:
 $[\text{Cu}(\text{NH}_3)_4]$ $[\text{PtCl}_4]$ and $[\text{Pt}(\text{NH}_3)_4]$ $[\text{CuCl}_4]$
- Write the name of the alloy used in Fischer Tropsch process in the synthesis of gasoline.
- Henry's law constant for $\text{CH}_3\text{Br}_{(g)}$ is $0.159 \text{ mol dm}^{-3} \text{ bar}^{-1}$ at 25°C . What is solubility of $\text{CH}_3\text{Br}_{(g)}$ in water at same temperature and partial pressure of 0.164 bar ?

SECTION – B

Attempt any EIGHT of the following questions:

[16]

- Explain pseudo-first order reaction with suitable example.
- Write the consequences of Schottky defect with reasons.
- What is the action of following on ethyl bromide:
 (i) Na in dry ether
 (ii) Mg in dry ether
- Explain formation of peptide linkage in protein with an example.
- Derive an expression to calculate molar mass of non volatile solute by osmotic pressure measurement.
- Explain monodentate and ambidentate ligands with example.
- Explain the trends in the following atomic properties of group 16 elements:
 (i) Atomic radii

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 - The Rise of Digital Reading 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - Advantages of eBooks Over Traditional Books
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 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
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 - Popular eBook Platforms
 - Features to Look for in an 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - User-Friendly Interface
4. Exploring eBook Recommendations from 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - Personalized Recommendations
 - 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board User Reviews and Ratings
 - 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board and Bestseller Lists
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 - ePub, PDF, MOBI, and More
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 - 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - Highlighting and Note-Taking 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - Interactive Elements 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
8. Staying Engaged with 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
9. Balancing eBooks and Physical Books 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - Setting Reading Goals 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - Fact-Checking eBook Content of 12th Board 2015 Mcq Answer For Chemistry Maharashtra Board
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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