# Ch 13 Study Guide (Chemical Bonds)

1. Fill in the table below with the correct values:

Element Name	lon Symbol	# Protons	# Electrons	Charge	Ion Type
Fluorine	F	9	10	-1	Anion
lodine	I-	53	54		Anien
Sulfar	52-	16	18	-2	Anion
Potassium	K+	19	18	+1	Cation
Calcium	Ca <sup>2+</sup>	20	18	+2	Carion
Browne	Bc-	35	36	-1	Anion
STRENTIAM	Sr2+	38	36	+2	Cation
Oxygen	03-	8	10	- 2	Anien
Magnesium	Ma2+	12	10	+2	Cation
Aluminum	Al 3+	13	10	+3	Carion
Selenium	Sea-	34	36	-2	Anion
Lithium	Li*	3	2	+1	Cation
Rubidium	Rb+	37	36	+1	Cation
Chlorine	ce-	17	18	-1	Anion

2. An ionic bond is formed between what two types of atoms?

Metal & non metal

3. A covalent bond is formed between what two types of atoms?

Nonmetal & Nonmetal

4. Draw the bond(s) that will form between Mg and O.

Mg - = 0: => Mg :0:

5. What is the formula for the compound formed in question #4?

# **13 Answer Key Chemistry**

John Bartsch

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