

Chemistry Worksheets Class 11 on Chapter 4 Chemical Bonding and Molecular Structure - Set 3

Q-1: Which of the fo	ollowing possesses an	expanded octet?
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- a) IF,
- b) SF.
- c) BrF_n
- d) All of the above

Q-2: Strongly electronegative element B contrasts with strongly electropositive element A. Both are divalent. The resulting compound would be

- a) A-B
- b) A2*B2-
- c) A2-B2"
- d) A-B

Q-3: Which of the following compounds shows the highest lattice energy?

- a) CsCl
- b) KCI
- c) RbCl
- d) NaCl

Q-4: Which one of the following molecules is formed by p-p overlapping?

- a) I_2
- b) HBr
- c) H₂S
- d) PH_a

Q-5: Which of the following H-bonds has the greatest impact on a molecule's physical characteristics?

- a) Intramolecular H-bond
- b) Intermolecular H-bond
- c) Both of the above
- d) All H-bonds

Q-6: State the crucial conditions that must be met for a molecule to undergo hybridisation.

Q-7: Describe why CH₄ has a tetrahedral geometry rather than a square planar geometry with a carbon atom in the centre and four H atoms at each corner.

Answer Key For Holt Science Chemical Bonding

R Barnett

Answer Key For Holt Science Chemical Bonding:

Unveiling the Energy of Verbal Beauty: An Psychological Sojourn through **Answer Key For Holt Science Chemical Bonding**

In a world inundated with displays and the cacophony of immediate transmission, the profound energy and psychological resonance of verbal beauty often fade in to obscurity, eclipsed by the continuous assault of sound and distractions. However, situated within the lyrical pages of **Answer Key For Holt Science Chemical Bonding**, a captivating perform of fictional brilliance that impulses with raw feelings, lies an unforgettable journey waiting to be embarked upon. Published by a virtuoso wordsmith, this mesmerizing opus guides visitors on a mental odyssey, delicately revealing the latent possible and profound affect embedded within the complex web of language. Within the heart-wrenching expanse of the evocative examination, we will embark upon an introspective exploration of the book is key subjects, dissect its fascinating publishing design, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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