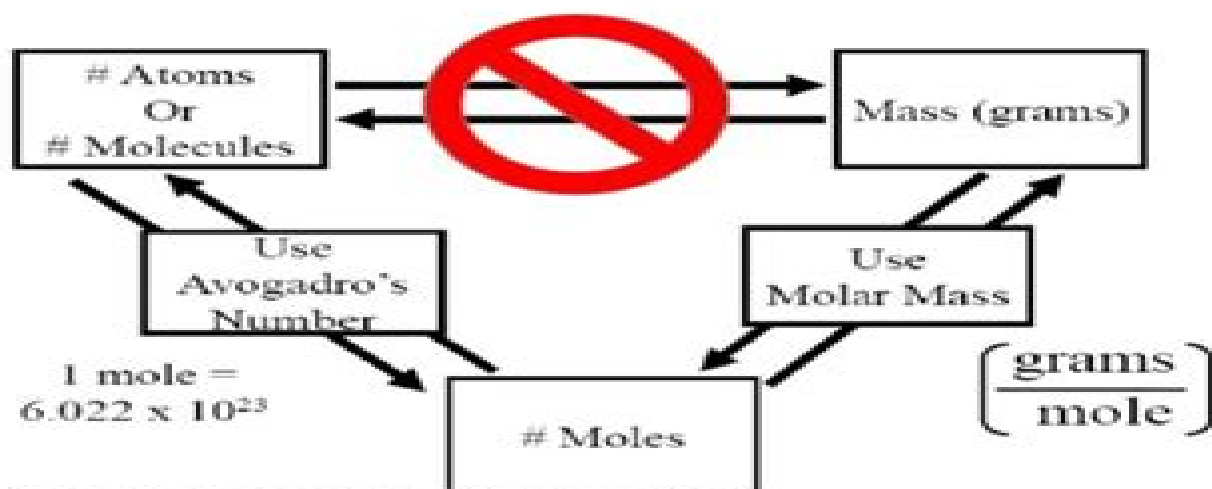


Moles, Molecules, and Grams Worksheet and Key



1) How many moles are there in 24.0 grams of FeF_3 ?

$$24 \div 19(3) + 56 = 0.212 \text{ mol}$$

2) How many moles are there in 458 grams of Na_2SO_4 ?

$$458 \div 23(2) + 32 + 16(4) = 3.23 \text{ mol}$$

3) How many grams are there in 2.30×10^{24} atoms of silver?

$$2.30 \times 10^{24} \div 6 \times 10^{23} = 3.83 \text{ mol}$$

$$3.83 = y \div 108$$

$$y = 414\text{g}$$

4) How many grams are there in 7.40 moles of AgNO_3 ?

$$7.40 = y \div 108 + 14 + 16 \times 3$$

$$7.40 = y \div 170$$

$$y = 1258\text{g}$$

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LP Steffe



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