- 6. What type of bond(s) were formed in the compound from question #4? Ionic Bond
- 7. Draw the bond(s) that will form between P and Cl.

:cecopooci: PCl3

8. What is the formula for the compound formed in question #7?

PCL3

9. What type of bond(s) were formed in the compound from question #7?

Covalent Bond

10. Draw the correct valence electron configuration for each atom given.

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. Ne	- Ge	B -	ĸ
ú	si -	- ċ -	Xe :

11. Fill in the table with the correct values.

Element	# valence electrons	# covalent bonds able to be formed
H	2	1
Не	2	0
С	4	4
CI	7	7
Ar	8	0
0	6	2
N	5	3
P	5	3

Answer Key For Chemistry Covalent Bond Notes Ch 8

Frank Brescia

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