



Amoeba Sisters | Video Recap

NAME: _____

More Properties of Water: Illustrated!

Complete the missing components of the table below!

Explain how illustration can relate to water!	Comic	Example of importance for living organisms?
5. 	Specific Heat 	6.
When frozen, water molecules are set into a lattice. The water molecules are less dense in ice form than in liquid form. Water expands when frozen and becomes less dense when it is frozen than liquid water.	7. Make your own illustration: 	8.
9.		Many organisms rely on this to cool them. The events in sweating are involved with this, but evaporative cooling is not isolated to animals! As mentioned in the video, water evaporating from plants can also result in cooling.

10. As mentioned throughout the video, **hydrogen bonds** are responsible for many of water's properties. Select one of the above properties and explain how hydrogen bonds contribute to the property.



Amoeba Sisters LLC



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Answer Key Of Water Aqueous Systems

TD Snyder



Answer Key Of Water Aqueous Systems:

Selected Water Resources Abstracts ,1988 **Selected Water Resources Abstracts** ,1988 Thermodynamics of Solutions Eli Ruckenstein,Ivan L. Shulgin,2009-06-17 This book consists of a number of papers regarding the thermodynamics and structure of multicomponent systems that we have published during the last decade Even though they involve different topics and different systems they have something in common which can be considered as the signature of the present book First these papers are concerned with difficult or very nonideal systems i e systems with very strong interactions e g hyd gen bonding between components or systems with large differences in the partial molar v umes of the components e g the aqueous solutions of proteins or systems that are far from normal conditions e g critical or near critical mixtures Second the conventional th modynamic methods are not sufficient for the accurate treatment of these mixtures Last but not least these systems are of interest for the pharmaceutical biomedical and related ind tries In order to meet the thermodynamic challenges involved in these complex mixtures we employed a variety of traditional methods but also new methods such as the fluctuation t ory of Kirkwood and Buff and ab initio quantum mechanical techniques The Kirkwood Buff KB theory is a rigorous formalism which is free of any of the proximations usually used in the thermodynamic treatment of multicomponent systems This theory appears to be very fruitful when applied to the above mentioned difficult systems

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