

MOLES AND MASS

Name _____

Determine the number of moles in each of the quantities below.

1. 25 g of NaCl

2. 125 g of H_2SO_4

3. 100. g of KMnO_4

4. 74 g of KCl

5. 35 g of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

Determine the number of grams in each of the quantities below.

1. 2.5 moles of NaCl

2. 0.50 moles of H_2SO_4

3. 1.70 moles of KMnO_4

4. 0.25 moles of KCl

5. 3.2 moles of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

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Nigel P. Freestone



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