

PHYSICAL SCIENCE PRACTICE TEST WITH ANSWER KEY

1. A weak attraction between hydrogen atom in one molecule and an oxygen in other called _____.
A. covalent bond
B. Hydrogen bond
C. Ionic bond
D. Metallic bond
2. Which of the following substances is most likely a covalent compound?
A. CaCO_3
B. KNO_3
C. SO_2
D. NaOH
3. What is the element needed to bond with Na to create the most ionic bond?
A. F
B. Cl
C. Br
D. I
4. The boiling point of substance A is higher than that of substance B. What does this mean?
A. Substance A is less viscous than substance B.
B. Substance A evaporates faster than substance B.
C. Substance B has greater surface tension than substance A.
D. Substance B has weaker intermolecular forces than substance A.
5. What is the intermolecular force of attraction between the nucleus of one atom and the electrons of a nearby atom?
A. Dipole-dipole force
B. Ion-dipole force
C. London dispersion force
D. Van der Waals force
6. Which of the following has the greatest number of atoms?
A. Ca(OH)_2
B. Al_2O_3

Answer Key For Physical Science

Responsive Education Solutions

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web terms in this set 11 a line on a map or chart along which there is a constant value isotherm a line on a map connecting points having the same temperature at a given time on average of a given period isobar a line on a map connecting points having the same atmospheric pressure at a given time on average of a given period

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web be interpreted as temperature therefore isotherm means same temperature isotherms drawn on temperature maps connect equal temperatures here are some helpful hints 1 isobars are drawn in 4 millibar intervals 2 always label the isobar value 3 pressures lower than the isobar value are always on one side of the isobar and higher