

Buoyancy Worksheet **Answers**

1) A medical ship has a mass of 100,000 kg. What volume of fresh water ($\rho = 1025 \text{ kg/m}^3$) will the ship displace?

$$100,000 \text{ [kg]} \div 1,025 \text{ [kg/m}^3\text{]} = 97,561 \text{ m}^3$$

2) How much mass can a 1,000,000 L balloon lift if the inside temperature of the balloon is 30 °C and the outside air temperature is 20 °C?

First, move the terms containing 'V' to the same side by subtracting 'V outside Fluid' from both sides:

$$M < V \text{ (outside fluid)} - V \text{ (inside fluid)}$$

Next, plug in the numbers:

$$M < (1,000,000 \text{ L}) \cdot (1.204 \text{ g/L} - 1.599 \text{ g/L})$$

$$M < 399 \text{ kg}$$

3) How many 10 L helium balloons would it take to lift a man in an armchair (75 kg) if the density of air is 1.2 g/L and the density of helium is 0.1786 g/L. Assume each balloon has a mass of 3 g. (Hint: First calculate how much extra mass a helium balloon can carry)

Just like the example in the lesson, one balloon can lift 0.2 g. If each balloon has a mass of 3 g, then each balloon can carry 7.2 g.

$$75 \text{ kg} \div 7.2 \text{ g} = 10,417 \text{ Balloons.}$$

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Table of Contents Answer Key Buoyancy

1. Understanding the eBook Answer Key Buoyancy
 - The Rise of Digital Reading Answer Key Buoyancy
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key Buoyancy
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key Buoyancy
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key Buoyancy
 - Personalized Recommendations
 - Answer Key Buoyancy User Reviews and Ratings
 - Answer Key Buoyancy and Bestseller Lists
5. Accessing Answer Key Buoyancy Free and Paid eBooks
 - Answer Key Buoyancy Public Domain eBooks
 - Answer Key Buoyancy eBook Subscription Services
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 - Answer Key Buoyancy Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key Buoyancy
 - Highlighting and Note-Taking Answer Key Buoyancy
 - Interactive Elements Answer Key Buoyancy
8. Staying Engaged with Answer Key Buoyancy

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Answer Key Buoyancy
- 9. Balancing eBooks and Physical Books Answer Key Buoyancy
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key Buoyancy
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key Buoyancy
 - Setting Reading Goals Answer Key Buoyancy
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key Buoyancy
 - Fact-Checking eBook Content of Answer Key Buoyancy
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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