

ABBREVIATIONS AND SYMBOLS			
amount of substance	<i>n</i>	joule	J
atmosphere	atm	kelvin	K
boiling pt. elev. const.	<i>K_b</i>	liter	L
Celsius temperature	°C	molar	<i>m</i>
energy of activation	<i>E_a</i>	molar	M
enthalpy	<i>H</i>	mole	mol
entropy	<i>S</i>	pressure	<i>P</i>
equilibrium constant	<i>K_{eq}</i>	second	s
free energy	<i>G</i>	specific heat capacity	<i>C_p</i>
freezing pt. depr. const.	<i>K_f</i>	temperature	<i>T</i>
gram	g	volt	V
hour	h	volume	V

CONSTANTS	
Planck's constant <i>h</i> = 6.626×10 ⁻³⁴ J·s	Gas Constant <i>R</i> = 0.0821 (L·atm)/(mol·K) <i>R</i> = 8.314 J/(mol·K)
Speed of light <i>c</i> = 2.998×10 ⁸ m·s ⁻¹	Water Dissociation Constant <i>K_w</i> = 1.0×10 ⁻¹⁴ at 25 °C
Rydberg constant <i>R_H</i> = 2.18×10 ⁻¹⁸ J	Faraday Constant <i>F</i> = 96470 C/mol e ⁻ <i>F</i> = 96470 A·s/mol e ⁻ <i>F</i> = 96470 J/V·mol e ⁻
Avogadro Constant <i>N_A</i> = 6.022×10 ²³ mol ⁻¹	
Measures of pressure 1 atm = 760 mmHg	

EQUATIONS			
$E = h\nu \quad E = \frac{hc}{\lambda}$	$\frac{\text{rate}_1}{\text{rate}_2} = \sqrt{\frac{\text{molar mass}_1}{\text{molar mass}_2}}$	Integrated Rate Laws zero: $[A] = [A]_0 - kt$	$\text{pH} = \text{p}K_a + \log\left(\frac{[A^-]}{[HA]}\right)$
$\Delta E = R_{\infty} \left(\frac{1}{n_f^2} - \frac{1}{n_i^2} \right)$	$\Delta T_1 = K_f m \quad \Delta T_2 = i K_f m$ $\Delta T_b = K_b m \quad \Delta T_b = i K_b m$ $\Pi = MRT \quad \Pi = i MRT$	second: $\frac{1}{[A]} = \frac{1}{[A]_0} + kt$	$\Delta G^\circ = \Delta H^\circ - T\Delta S^\circ$
$q = mc\Delta T$ $PV = nRT$	$\ln\left(\frac{k_2}{k_1}\right) = \frac{E_a}{R} \left(\frac{1}{T_1} - \frac{1}{T_2} \right)$	first: $\ln[A] = \ln[A]_0 - kt$	$\Delta G^\circ = -RT \ln K \quad \Delta G = \Delta G^\circ + RT \ln Q$ $E^\circ = \frac{RT}{nF} \ln K \quad E = E^\circ - \frac{RT}{nF} \ln Q$

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																		58 Ce 140.1	59 Pr 140.9	60 Nd 144.2	61 Pm 144.9	62 Sm 150.4	63 Eu 152.0	64 Gd 157.3	65 Tb 158.9	66 Dy 162.5	67 Ho 164.9	68 Er 167.3	69 Tm 168.9	70 Yb 173.0	71 Lu 175.0
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Answer Key For Acs Divched Examination Institute

Thomas G. Waddell, Thomas R. Rybolt

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