

Analytical Chemistry MCQs: TYBSC SEM III

8. The solubility product K_{sp} of sparingly soluble Ag_2CrO_4 is 4×10^{-12} . The molar solubility of the salt is
- (a) 1.0×10^{-4} moles/lit
 - (b) 2.0×10^{-4} moles/lit
 - (c) 1.0×10^{-5} moles/lit
 - (d) 1.2×10^{-12} moles/lit
9. Solubility product of $AgCl$ is 1.44×10^{-4} moles/lit at 373K. The solubility of the $AgCl$ in boiling of water will be....
- (a) 0.72×10^{-4} moles/lit
 - (b) 0.72×10^{-2} moles/lit
 - (c) 1.20×10^{-2} moles/lit
 - (d) 1.20×10^{-4} moles/lit
10. Precipitation takes place when the ionic product....
- (a) Is zero
 - (b) Exceeds the solubility product
 - (c) Equal to solubility product
 - (d) Is less than the solubility product
11. The electromotive force is expressed in the unit of
- (a) Ohm
 - (b) Volt
 - (c) Ampere
 - (d) Coulomb
12. If the cell can provide electrical energy to an external system, it is called as
- (a) Galvanic cell
 - (b) Electrolytic cell
 - (c) Concentration cell
 - (d) Chemical cell
13. The current per unit area of electrode surface is known as....
- (a) Current efficiency
 - (b) Current yield
 - (c) Current density
 - (d) None of these

Analytical Chemistry Questions And Answers

PT Brinkman



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Quality Control in Analytical Chemistry G. Kateman, L. Buydens, 1993-09-06 Describes the basics of analytical techniques sampling and data handling in order to improve quality control in analytical laboratory management Stresses what quality parameters can be improved and which ones should be rectified first This edition includes numerous modern methods and the latest developments in time proven techniques **Basic Analytical Chemistry** Mr. Rohit

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Valcarcel, 2012-12-06 Principles of Analytical Chemistry aims to ease the first contact of students pursuing different scientific and technical studies by providing them with a simple general overview of the discipline The objective of this innovative textbook is to teach rather than to inform Using keywords of modern Analytical Chemistry it constructs a teaching message accessible to the beginner with emphasis on relations between these keywords to ensure consistency in the teaching contents of the book Practical exercises 450 questions for students and 27 interactive seminars are used as vehicles for expounding a large number of examples in order to clarify and consolidate the text and facilitate its comprehension Analytical

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Analytical Chemistry Victor Angelo Soffiantini, 2021-10-11 The textbook is based on the APPLIED use of laboratory instrumentation and apparatus in practice in the real working world with absolute minimum use of complex calculations and mathematics Instrumental theory is kept to a minimum with useful practical hints and unbiased instruction on lab instruments capabilities and operations All text is in simple to understand language of the complexities of chemical analyses

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