Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



- What type of charge does a proton have? Positive (+1)
- What type of charge does a neutron have? Neutral (0)
- What type of charge does an electron have? Negative (-1)
- Which two subatomic particles are located in the nucleus of an atom?
 Protons and neutrons
- 8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
 35
- 9. What is the atomic number of the atom in the diagram above?
- What is the atomic mass/mass number of the atom in the diagram above?
 (11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number
- 11. How many protons are in the nucleus of an atom with an atomic number of 15?
 15 *number of protons is the same as the atomic number
- How many electrons are in the nucleus of an atom with an atomic number of 20?
- How many neutrons are in the nucleus of an atom with an atomic number of 25? (use Periodic Table for mass)
 Neutrons, atomic mass: 54.938 ~ 55 atomic number: 25....55-25=30
- 14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
 Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.
- How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
 Neutrons= 11 (atomic mass: 36 - 25 :atomic number)

Answers For Atomic Structure

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Answers For Atomic Structure:

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In a digitally-driven earth wherever displays reign great and quick conversation drowns out the subtleties of language, the profound techniques and psychological nuances hidden within words usually move unheard. Yet, located within the pages of **Answers For Atomic Structure** a charming literary prize blinking with fresh thoughts, lies a fantastic quest waiting to be undertaken. Penned by an experienced wordsmith, that charming opus encourages readers on an introspective journey, gently unraveling the veiled truths and profound influence resonating within ab muscles cloth of each and every word. Within the psychological depths with this moving review, we will embark upon a genuine exploration of the book is key subjects, dissect their charming writing design, and succumb to the strong resonance it evokes strong within the recesses of readers hearts.

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