# Chemical Formula Writing Worksheet Solutions

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl<sub>2</sub>", as shown.

Service	zinc	iron (II)	iron (III)	gallium	silver	lead (IV)
chloride	ZnCl <sub>2</sub>	FeCl <sub>2</sub>	FeCl <sub>3</sub>	GaCl <sub>3</sub>	AgCI	PbCI <sub>4</sub>
acetate	Zn(C <sub>2</sub> H <sub>1</sub> O <sub>2</sub> ) <sub>2</sub>	Fe(C <sub>2</sub> H <sub>2</sub> O <sub>2</sub> ) <sub>2</sub>	Fe(C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> ) <sub>3</sub>	Ga(C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> ) <sub>3</sub>	Ag C <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	Pb(C <sub>2</sub> H <sub>1</sub> O <sub>2</sub> ) <sub>4</sub>
nitrate	Zn(NO <sub>3</sub> ) <sub>2</sub>	Fe(NO <sub>3</sub> ) <sub>2</sub>	Fe(NO <sub>3</sub> ) <sub>3</sub>	Ga(NO <sub>3</sub> ) <sub>3</sub>	AgNO <sub>3</sub>	Pb(NO <sub>3</sub> ) <sub>4</sub>
oxide	ZnO	FeO	Fe <sub>2</sub> O <sub>3</sub>	Ga <sub>2</sub> O <sub>3</sub>	Ag <sub>2</sub> O	PbO <sub>2</sub>
nitride	Zn <sub>3</sub> N <sub>2</sub>	Fe <sub>3</sub> N <sub>2</sub>	FeN	GaN	Ag <sub>2</sub> N	Pb <sub>3</sub> N <sub>4</sub>
sulfate	ZnSO <sub>4</sub>	FeSO <sub>4</sub>	Fe <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	Ga <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	Ag <sub>2</sub> SO <sub>4</sub>	Pb(SO <sub>4</sub> ) <sub>2</sub>

Write the formulas for the following compounds:

- copper (II) chloride CuCl<sub>2</sub>
- lithium acetate LiC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>
- vanadium (III) selenide V<sub>2</sub>Se<sub>3</sub>
- 4) manganese (IV) nitride Mn<sub>3</sub>N<sub>4</sub>
- 5) beryllium oxide BeO
- sodium sulfate Na<sub>2</sub>SO<sub>4</sub>
- 7) aluminum arsenide AlAs
- potassium permanganate KMnO<sub>4</sub>
- 9) chromium (VI) cyanide Cr(CN)<sub>6</sub>
- 10) tin (II) sulfite SnSO<sub>3</sub>
- 11) vanadium (V) fluoride VF<sub>5</sub>
- ammonium nitrate NH<sub>4</sub>NO<sub>3</sub>

# **Answers For Ionic Formula Writing**

**R.D.MADAN** 

#### **Answers For Ionic Formula Writing:**

<u>Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice</u> Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover

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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order A New Approach to ICSE Chemistry for Class X (A.Y. 2023-24)Onward V. K. Sally, 2023-05-20 A New Approach to ICSE Chemistry for Class X has been revised in accordance with the latest syllabus prescribed by the Council for the Indian School Certificate Examination New Delhi for Class X The main strength of the book lies in the presentation of scientific content which has been so arranged that the topics are linked with each other and do not cause any unnecessary burden on the mind of the student Salient features of this book are as follows Clear simple and easy language A large number of chemical reactions along with experimental observations Full sized diagrams as expected from a student in the examination Topic wise Video Lectures as a support for effective

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