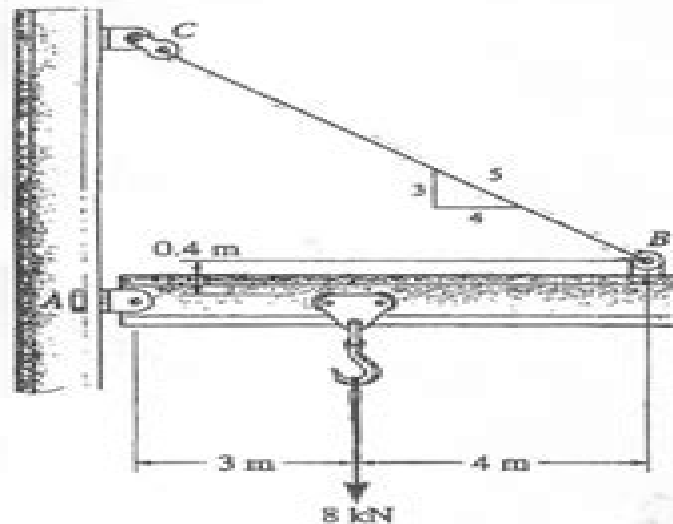


General Instructions

1. Closed books and closed notes. A one side A4 size formula sheet is allowed.
2. Write all your solutions in the space provided. Using back of the sheets is ok but you need clearly indicated on the front sheets.
3. You must draw a FBD for all the equilibrium problems. Make the diagrams and sketches as clear as possible, and show your derivations clearly. Incomplete solutions will receive only partial credit even if the answer is correct. Present your results in three significant digits with proper units; box your final results.

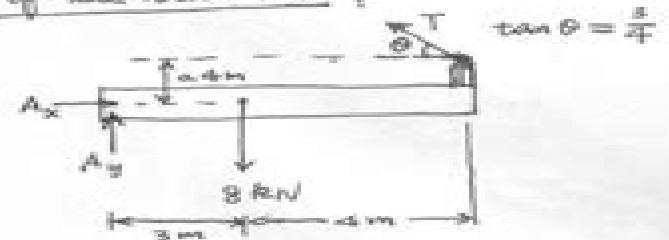
Problem 1 (20 points): A jib crane AB is shown below. It is pin-connected at A and supported by a cable BC. Neglect its weight. Find the reaction force at A and the cable tension.



Knowns: pin support at A;
BC is a cable; neglect the ^{beam} weight.

Find: reaction force at A: A_x, A_y ;
cable tension, T .

FBD of the beam AB:



Eqs. of Equilibrium:

$$(\Sigma) \Sigma M_A = 0: (T \cos \theta)(0.4m) + (T \sin \theta)(7m) - (8 \text{ kN})(3m) = 0$$

$$\text{Hence, } T \left(\frac{3}{4} \right)(0.4m) + T \left(\frac{3}{5} \right)(7m) = 24 \text{ kN} \cdot m$$

$$\boxed{T = 5.31 \text{ kN}}$$

$$\text{Also, } (\Sigma) \Sigma F_x = 0: A_x - T \left(\frac{3}{5} \right) = 0, \quad A_x = \frac{3}{5} T = \boxed{4.25 \text{ kN} = A_x}$$

$$(\Sigma) \Sigma F_y = 0: A_y - 8 \text{ kN} + T \left(\frac{4}{5} \right) = 0, \quad A_y = 8 \text{ kN} - \frac{4}{5} (5.31) = \boxed{4.81 \text{ kN} = A_y}$$

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Lei Shi

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k

for temperature into the ideal gas equation and solve for r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts

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