

Data Report Sheet: Analysis of an Unknown Chloride

Unknown # 32

Molarity of Standard $\text{Ag}(\text{NO}_3)_3$ 0.1482 Moles/liter

	Trial I	Trial II	Trail III
Mass of Unknown	0.3991 g	0.3991 g	0.3991 g
Initial Burette Reading	49.3	29.5	37.6 ML
Final Burette Reading	29.5	9.8	17.8ML
Vol. $\text{Ag}(\text{NO}_3)$ Added	?	?	?
Moles $\text{Ag}(\text{NO}_3)$	?	?	?
Moles Cl^-	?	?	?
Mass Cl^-	?	?	?
% Cl^- in Unknown	?	?	?

Average % Cl^- in unknown ? %

Analysis Of An Unknown Chloride Answers

J Spring



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Environmental Sampling and Analysis Maria Csuros,1997-03-24 This manual covers the latest laboratory techniques state of the art instrumentation laboratory safety and quality assurance and quality control requirements In addition to complete coverage of laboratory techniques it also provides an introduction to the inorganic nonmetallic constituents in environmental samples their chemistry and their control by regulations and standards Environmental Sampling and Analysis Laboratory Manual is perfect for college and graduate students learning laboratory practices as well as consultants and regulators who make evaluations and quality control decisions Anyone performing laboratory procedures in an environmental lab will appreciate this unique and valuable text , Plating ,1950 **Cambridge IGCSE® Chemistry Practical**

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Dasgupta,Kevin A. Schug,2013-10-07 With the 7th Edition of Analytical Chemistry renowned chemists Purnendu Sandy Dasgupta and Kevin Schug both of the University of Texas Arlington join the author team The new edition focuses on more in depth coverage of the principles and techniques of quantitative analysis and instrumental analysis aka Analytical Chemistry The goal of the text is to provide a foundation of the analytical process tools and computational methods and resources and to illustrate with problems that bring realism to the practice and importance of analytical chemistry It is designed for undergraduate college students majoring in chemistry and in fields related to chemistry **First Course in Quantitative**

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