

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*
S *sulphur*
Cl *chlorine*
C *carbon*
N *nitrogen*
Na *sodium*
Pb *lead*
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*
phosphorus *P*
fluorine *F*
magnesium *Mg*
potassium *K*
calcium *Ca*
zinc *Zn*
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79** copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5** calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Lead	82	207	82	125	82
Argon	18	58	18	40	18
Barium	56	137	56	61	56
Tin	50	112	50	62	50

6. Match the unit with its description:

F 1. amount of substance
C 2. density
B 3. mass
D 4. molar mass
A 5. energy
E 6. volume

A. joule
B. g
C. g/mL
D. g/mole
E. L
F. mole

Answers For Merrill Chemistry Review

RD Boyd



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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 Securities and Exchange Commission Staff Report on Transactions in Securities of the City of New York Finance U.S. Congress. Senate. Committee on Banking (and Urban Affairs), United States. Securities and Exchange Commission, 1977 *Securities and Exchange Commission Staff Report on Transactions in Securities of the City of New York* United States. Securities and Exchange Commission, 1977

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Table of Contents Answers For Merrill Chemistry Review

1. Understanding the eBook Answers For Merrill Chemistry Review
 - The Rise of Digital Reading Answers For Merrill Chemistry Review
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Merrill Chemistry Review
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Merrill Chemistry Review
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Merrill Chemistry Review
 - Personalized Recommendations
 - Answers For Merrill Chemistry Review User Reviews and Ratings
 - Answers For Merrill Chemistry Review and Bestseller Lists
5. Accessing Answers For Merrill Chemistry Review Free and Paid eBooks

- Answers For Merrill Chemistry Review Public Domain eBooks
- Answers For Merrill Chemistry Review eBook Subscription Services
- Answers For Merrill Chemistry Review Budget-Friendly Options
- 6. Navigating Answers For Merrill Chemistry Review eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Merrill Chemistry Review Compatibility with Devices
 - Answers For Merrill Chemistry Review Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Merrill Chemistry Review
 - Highlighting and Note-Taking Answers For Merrill Chemistry Review
 - Interactive Elements Answers For Merrill Chemistry Review
- 8. Staying Engaged with Answers For Merrill Chemistry Review
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Merrill Chemistry Review
- 9. Balancing eBooks and Physical Books Answers For Merrill Chemistry Review
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Merrill Chemistry Review
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Merrill Chemistry Review
 - Setting Reading Goals Answers For Merrill Chemistry Review
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Merrill Chemistry Review
 - Fact-Checking eBook Content of Answers For Merrill Chemistry Review
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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