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# Assignment #2 – Compound Names and Formulas Multi-valent ions only

| A. Write the correct formula for the | e following compounds, all of which have been named using the         |
|--------------------------------------|---|
|                                      | The ionic charge is given after the first element (metallic) in Roman |
| Numerals.                            |   |

| 1. copper (II) oxide CUO   | 11. manganese (III) oxide Mv 203    |  |  |
|--|-------------------------------------|--|--|
| 2. mercury (I) oxide HQ20  | 12. vanadium (II) bromide VB(z      |  |  |
| 3. gold (III) chloride _AuCl3  | 13. niobium(V) oxide Nb205          |  |  |
| 4. thallium (III) bromide TIPY 3   | 14. titanium (III) oxide            |  |  |
| 5. bismuth (V) oxide   | 15. titanium (III) nitride TIN      |  |  |
| 6. terbium (III) oxide   | 16. iron (II) oxide <u>Fe.O</u>     |  |  |
| 7. uranium (VI) oxide \( \lambda \ \O \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \            | 17. cobalt (II) phosphide C03P2     |  |  |
| 8. protactinium (IV) bromide PhBY4   | 18. tin (II) oxide                  |  |  |
| 9. cerium (III) oxide  | 19. thulium (II) bromide TWBY 2     |  |  |
| 10. arsenic (V) sulphideAS_2S_5  | 20. copper (II) bromide CUPY2       |  |  |
|  |                                     |  |  |
| B. Name the compound using the Roman Numeral Method and balance the ionic charges. |                                     |  |  |
| 1. SnCl Tin (IV) Chloride  | 9. Pdf. Palladium (IV) Fluoride     |  |  |
| 2. BIBIS BISMINTO STOMICE  | 10. OszOs OSMIUM (III) Oxide        |  |  |
| 3. POO2 ROLONIUM (IV) OXICE  | 11. MOBEZ MOLYBORENIUM (II) Bramide |  |  |
| 4. Pbl2 100d(II) Iodide  | 12. VCIs Vanadium (V) Chloride      |  |  |
| 5. HgO <u>Mercury</u> (II) Oxide   | 13. Mn2O3 Manganese (III) Oxide     |  |  |
| 6. HgCI MEYCUYY (I) Chroride   | 14. COO COLONY (II) Oxide           |  |  |
| 7. Auzos Gold (III) Oxide  | 15. Np2O3 NEDTUNIUM (II) OXICE      |  |  |
| 8. FeClo Iron (II) Chloride  | 16. V2O6 Vanadium (V) Oxide         |  |  |

# **Answers For Writing Formulas From Names**

Neil D. Jespersen, Alison Hyslop

#### **Answers For Writing Formulas From Names:**

Chemical formulas and names Virginia P. Powell, 1965 Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students A Textbook of Organic Chemistry, 22e Arun Bahl & B S Bahl, 2019 With an increased focus on fundamentals this new edition of A Textbook of Organic Chemistry continues to present the time tested functional group approach to the subject This examination oriented book breaks the intricacies of Organic Chemistry into easy to understand steps which gives the student the necessary foundation to build upon learn and understand Organic Chemistry in a way that is efficient as well as long lasting Foundations of College Chemistry Morris Hein, Susan Arena, 2013-01-01 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years Foundations of College Chemistry Alternate 14th Edition has helped readers master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis A Textbook of Organic Chemistry, 23e Arun Bahl & B S Bahl, A Textbook of Organic Chemistry has been a market leader for more than 75 years This examination oriented book breaks the intricacies of organic chemistry into easy to understand steps which gives the student the necessary foundation to build upon learn and ultimately understand organic chemistry. The textbook has been written with the student in mind The

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