



All answers should go inside the **RED BOXES**.

Name: _____

Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these **BEFORE** using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers?

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have?

2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**. What happens?

3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**. What happens now?



Answers Ofexploration Guide Covalent Bonds

Andrew Allott



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Table of Contents Answers Ofexploration Guide Covalent Bonds

1. Understanding the eBook Answers Ofexploration Guide Covalent Bonds
 - The Rise of Digital Reading Answers Ofexploration Guide Covalent Bonds
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers Ofexploration Guide Covalent Bonds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers Ofexploration Guide Covalent Bonds
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers Ofexploration Guide Covalent Bonds
 - Personalized Recommendations
 - Answers Ofexploration Guide Covalent Bonds User Reviews and Ratings
 - Answers Ofexploration Guide Covalent Bonds and Bestseller Lists
5. Accessing Answers Ofexploration Guide Covalent Bonds Free and Paid eBooks

- Answers Ofexploration Guide Covalent Bonds Public Domain eBooks
 - Answers Ofexploration Guide Covalent Bonds eBook Subscription Services
 - Answers Ofexploration Guide Covalent Bonds Budget-Friendly Options
6. Navigating Answers Ofexploration Guide Covalent Bonds eBook Formats
- ePub, PDF, MOBI, and More
 - Answers Ofexploration Guide Covalent Bonds Compatibility with Devices
 - Answers Ofexploration Guide Covalent Bonds Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Answers Ofexploration Guide Covalent Bonds
 - Highlighting and Note-Taking Answers Ofexploration Guide Covalent Bonds
 - Interactive Elements Answers Ofexploration Guide Covalent Bonds
8. Staying Engaged with Answers Ofexploration Guide Covalent Bonds
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers Ofexploration Guide Covalent Bonds
9. Balancing eBooks and Physical Books Answers Ofexploration Guide Covalent Bonds
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers Ofexploration Guide Covalent Bonds
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers Ofexploration Guide Covalent Bonds
- Setting Reading Goals Answers Ofexploration Guide Covalent Bonds
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers Ofexploration Guide Covalent Bonds
- Fact-Checking eBook Content of Answers Ofexploration Guide Covalent Bonds
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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