

7.2 PhET LAB: Alpha Decay



Name: _____

OBJECTIVE: explain the alpha decay process; complete radioactive decay equations; define and analyze half-life through applying the PhET "Alpha Decay" simulation at <http://phet.colorado.edu>.

Open/Run Alpha Decay at <http://phet.colorado.edu>.

Take 5 minutes to freely explore the sim.

Investigating Alpha Decay

1. Start on the **SINGLE ATOM** tab. Observe the decay of Polonium-211. After each decay, press the RESET NUCLEUS button to watch the process again. Write a description of alpha decay for Po-211.

Complete the following alpha decay equations, using <http://www.chemicalelements.com> as a resource:



7. How is alpha decay used in everyday life? (give at least two uses)

Investigating Half-Life of Alpha Decay

8. Click the **MULTIPLE ATOMS** tab. Execute five trials to determine the number of parent and daughter nuclei at one half-life. Complete the table below.

Parent Nuclei Po-211	Parent Nuclei (remaining nuclei) [yellow]	Daughter Nuclei (decayed nuclei) [black]
100		
80		
60		
40		
20		

9. Define half-life.

Alpha Decay Phet Lab Answers

A. Săndulescu, M. Rizea, I. Târnoveanu



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Theory of Alpha Decay W. Tobocman, 1962 *Alpha Decay Widths and the Structure of the Alpha Particle* A. Săndulescu, M. Rizea, I. Târnoveanu, 1977 Alpha Decay from Deformed Nuclei Tore Berggren, Peter Olanders, 1987
Contributions to the Theory of Alpha Decay Per Olof Fröman, 1957 **Alpha Decay in Isotopes of Atomic Number Less Than 83**, 1949 Some time ago we started work in an attempt to observe alpha particle decay in isotopes of atomic number less than 83 In the first experiments thin targets of gold leaf were bombarded with 190 MeV deuterons in the 184 inch cyclotron Two alpha decay periods were observed in these targets one of 0.7 minutes half life and another of 4.3 minutes half life The alpha particle energies were 5.7 and 5.2 MeV respectively Chemical separations proved that the 4.3 minute period is due to a gold isotope and suggested that the 0.7 minute period is due to a mercury isotope The mass numbers of these new isotopes have not been determined However the results of excitation functions in the production of the gold isotope by bombarding gold and platinum with protons suggest that its mass number lies in the range 185-188 The work on this isotope indicates that the alpha to electron capture branching ratio is of the order of magnitude of 10⁻⁴ and that positron activity accompanies the 4.3 minute alpha period **Alpha Decay as a Weak Interaction** A. O. Barut, A. Sevgen, 1973
Alpha Decay from Deformed Nuclei Tore Berggren, Peter Olanders, 1987 **Alpha Decay as a Fission-like Process** D. N. Poenaru, M. Ivascu, A. Sandulescu, 1979 **Alpha Decay and Excited States of ¹¹¹Xe** Luigi Capponi, 2014 **On the First Structure in Alpha Decay** Age Bohr, P.O. Fröman, Ben R. Mottelson, 1955

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