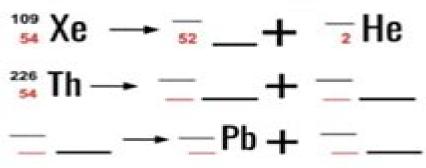
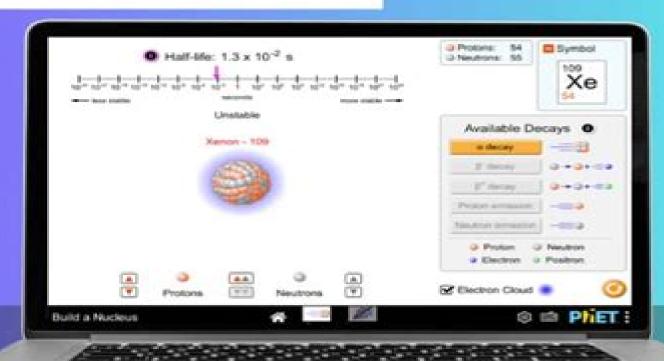
Students will practice making equations for alpha and beta decay in order to deepen understanding.

6. Complete the equations below, which display the process of alpha decay.





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Per Olof Fröman

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Theory of Alpha Decay W. Tobocman, 1962 Alpha Decay Widths and the Structure of the Alpha Particle A. Săndulescu, M. Rizea, I. Târnoveanu, 1977 Contributions to the Theory of Alpha Decay Per Olof Fröman, 1957 Decay from Deformed Nuclei Tore Berggren, Peter Olanders, 1987 On the Theory of Alpha Decay Pieter Johannes Alpha Decay as a Weak Interaction A. O. Barut, A. Sevgen, 1973 Alpha Decay in Isotopes of Atomic Brussaard, 1958 Number Less Than 83,1949 Some time ago we started work in an attempt to observe alpha particle decay in isotopes of atomic number less than 83 In the first experiments thin targets of gold leaf were bombarded with 190 Mev deuterons in the 184 inch cyclotron Two alpha decay periods were observed in these targets one of 0.7 minutes half life and another of 4.3 minutes half life The alpha particle energies were 5 7 and 5 2 Mev respectively Chemical separations proved that the 4 3 minute period is due to a gold isotope and suggested that the 0.7 minute period is due to a mercury isotope. The mass numbers of these new isotopes have not been determined However the results of excitation functions in the production of the gold isotope by bombarding gold and platinum with protons suggest that its mass number lies in the range 185 188 The work on this isotope indicates that the alpha to electron capture branching ratio is of the order of magnitude of 10 4 and that positron activity accompanies the 4 3 minute alpha period Alpha Decay as a Fission-like Process D. N. Poenaru, M. Ivascu, A. Sandulescu, 1979 Alpha Decay and Excited States of 111Xe Luigi Capponi,2014 **Alpha Decay from Deformed Nuclei** Tore Berggren, Peter Olanders, 1987

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