

# CHM 111 Analysis of Aluminum-Zinc Alloy using Ideal Gas Law

## Report Sheet

### Data Sheet:

|                                          | Trial 1                               | Trial 2                                                            |
|------------------------------------------|---------------------------------------|--------------------------------------------------------------------|
| Mass of gelatin capsule                  | <u>1.7382</u> g                       | <u>/</u> g                                                         |
| Mass of alloy sample plus capsule        | <u>1.9852</u> g                       | <u>/</u> g                                                         |
| Mass of empty beaker                     | <u>104.3024</u> g                     | <u>/</u> g                                                         |
| Mass of beaker plus displaced water      | <u>346.225</u> g                      | <u>/</u> g                                                         |
| Barometric pressure                      | <u>                    </u> mm Hg     | <u>                    </u> mm Hg                                  |
| Temperature                              | <u>24</u> °C                          | <u>                    </u> °C                                     |
| Calculations Mass of alloy sample        | <u>0.247</u> g                        | <u>/</u> g                                                         |
| Mass of displaced water                  | <u>241.9226</u> g                     | <u>/</u> g                                                         |
| Volume of displaced water                | <u>                    </u> mL        | <u>/</u> mL                                                        |
| Volume of H <sub>2</sub> , V             | <u>                    </u> L         | <u>/</u> L                                                         |
| Temperature of H <sub>2</sub> , T        | <u>297.15</u> K                       | <u>                    </u> K                                      |
| Vapor pressure of water at T             | <u>22.4</u> mm Hg                     | <u>                    </u> mm Hg                                  |
| Pressure of dry H <sub>2</sub> ,         | <u>755.7</u> mm Hg; <u>0.9943</u> atm | <u>                    </u> mm Hg; <u>                    </u> atm |
| Moles H <sub>2</sub> from sample,        | <u>                    </u> moles     | <u>/</u> moles                                                     |
| Moles H <sub>2</sub> per gram of sample, | <u>                    </u> moles/g   | <u>/</u> moles/g                                                   |
| Al (read from graph)                     | <u>                    </u> %         | <u>/</u> %                                                         |
| Unknown no. <u>3</u>                     |                                       |                                                                    |

# Analysis Of Aluminum Zinc Alloy Lab Answers

**AW Chickering**



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