# **Buffers Keep the Balance**

ata Table 1. Model Carbonate Blood Buffer

Table 1	Test Tube Number						
	1	2	3	4	5	6	
Solution	Carbonic Acid (Reference)	Model Carbonate Blood Buffer	Water (Control)	Model Carbonate Blood Buffer	Water (Control)	Sodium Bicarbonate (Reference)	
Bromthymol blue indicator color	Yenow GED	green 8h	green 600	men 13h	9 Heen 600	bive	
Initial pH value	6.0	7.2	6.0	7.0	6.0	8.0	
Number of drops of HCl required to convert solutions 2 and 3 to yellow (acid reference color)	NA	16	5	NA	NA	NA	
Number of drops of NaOH required to convert solutions 4 and 5 to blue "hase reference color)	NA	NA	NA	2	2	NA	

# Data Table 2. Effect of HCl on Biological Phosphate Buffers

	Estimated pH (pH paper)	Universal Indicator Color				
		Initial	After I drop of HCI	After 5 drops of HCI	After 10 drops of HCl	
Water (control)	6.0	6.0	6.0	NA	NA.	
Phosphate Buffer A	6.0	60	6.0	6.0	10	
Phosphate Buffer B	6.4	6.4	6.4	6.0	6.0	

# Data Table 3. Effect of NaOH on Biological Phosphate Buffers

	-	Universal Indicator Color				
	(pH paper)	Initial	After 1 drop of NaOH	After 5 drops of NaOH	After 10 drops of HCI	
(control)	6.0	6.0	7.0	NA	NA NA	
Phosphate Buffer A	6.0	6.0	6.3	9 .		
Phosphate Buffer B	6.4	6.4	8.0	9.0	8.0	

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Puja Mehta

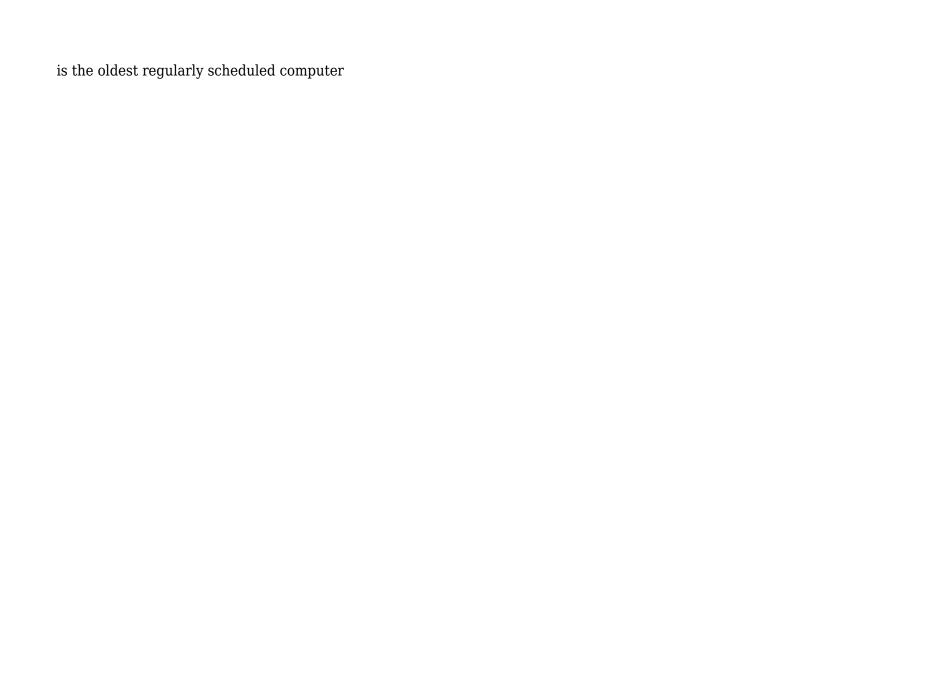
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