UNIVERSITY OF EDUCATION

"UExam" Semester-VII, 2024 BS Chemistry, Session: 2020-24

Course Code: CHEM4111

Subject: Inorganic Reaction Mechanism

Time Allowed: 100 Minutes.

Max. Marks: 42

Section II (Short Answer)

Q.2- Write short answers of the following.

9x2 = 18

- : What is the relationship between polarisability and trans-effect?
- N How metal-ligand pi-bond form in the ligands of Trans directing series lying at the high end?
- 26 Explain the conditions of reductive elimination reaction?
- What are the success and weakness of electrostatic polarization theory?
- What is Kumakov's test?
- What are the conditions for oxidative reactions?
- √ii Draw and explain Trans directing series.
- →W¹ Compare S_N¹ and S_N² mechanism in octahedral substitution reactions.
- ix In which reactions of octahedral complexes, the Trans effect become more important?

Section III (Essay Type)

Answer the following Questions

3x8 = 24

Q. #.3.

(8 Marks)

Derive the relationship $k_{obs} = k_o + k_V$ [Y'] by considering substitution reactions in Pt(II) square planar complexes and also explain these are stereospecific reactions.

Q. #.4.

(8 Marks)

Discuss the mechanism of acid hydrolysis by taking the example of octahedral complex of Co(III). Also explain in detail the factors which affect the rate constant of acid hydrolysis reactions.

Q. #.5.

(5+3=8 Marks)

- a) Explain in detail the factors affecting reductive elimination reaction with appropriate reactions.
- b) Justify, "Reductive elimination is the reverse of oxidative addition".

14 F321 Past Paper

Lei Huang

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