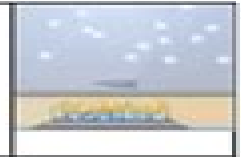


Activity C:

Altitude and phase changes

Get the Gizmo ready:

- Click **Reset**.
- Set **Ice volume** to 0 cm³.
- Set the **Altitude** to 5,000 meters (16,404 feet).



Question: The **altitude** of a location is its vertical distance above sea level. How does altitude affect phase changes?

1. **Form a hypothesis:** As altitude increases, the air pressure decreases. How do you think the lower pressure will affect the following? (Highlight your answers. ✎)

A. Freezing point:	Increase	Stay the same	Decrease
B. Melting point:	Increase	Stay the same	Decrease
C. Boiling point:	Increase	Stay the same	Decrease

2. **Experiment:** Use the Gizmo to find the freezing, melting, and boiling points of water at 5,000 meters (16,404 feet). Write these values below.

Freezing point: Melting point: Boiling point:

3. **Analyze:** How did altitude affect the freezing, melting, and boiling points of water?

Boiling point dropped, while freezing and melting points remained unchanged.

4. **Challenge:** Try to explain these results based on the fact that air pressure decreases with altitude. If possible, discuss your ideas with your classmates and teacher.

Because there is less pressure on it, breaking the bonds requires less kinetic energy when there is less air pressure.

5. **Apply:** Why does pasta take longer to cook in the mountains?

The pasta cooks more slowly because the water can't reach a high enough temperature.

6. **Apply:** A pressure cooker allows food to be cooked under high pressure. Why is this useful?

It cooks more quickly under increased pressure because it reaches a higher temperature.

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S Baum



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