

◀ Homework Chapter 6

Exercise 6.39: Problems - The Mole Concept

Part A

How many carbon atoms are in a diamond (pure carbon) with a mass of 75 mg?

Express your answer using two significant figures.

$$N = 2.5 \cdot 10^{21} \quad \text{C atoms}$$

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Table of Contents Answers Mastering Chemistry Homework

1. Understanding the eBook Answers Mastering Chemistry Homework
 - The Rise of Digital Reading Answers Mastering Chemistry Homework
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers Mastering Chemistry Homework
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers Mastering Chemistry Homework
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers Mastering Chemistry Homework
 - Personalized Recommendations

- Answers Mastering Chemistry Homework User Reviews and Ratings
- Answers Mastering Chemistry Homework and Bestseller Lists
- 5. Accessing Answers Mastering Chemistry Homework Free and Paid eBooks
 - Answers Mastering Chemistry Homework Public Domain eBooks
 - Answers Mastering Chemistry Homework eBook Subscription Services
 - Answers Mastering Chemistry Homework Budget-Friendly Options
- 6. Navigating Answers Mastering Chemistry Homework eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers Mastering Chemistry Homework Compatibility with Devices
 - Answers Mastering Chemistry Homework Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers Mastering Chemistry Homework
 - Highlighting and Note-Taking Answers Mastering Chemistry Homework
 - Interactive Elements Answers Mastering Chemistry Homework
- 8. Staying Engaged with Answers Mastering Chemistry Homework
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers Mastering Chemistry Homework
- 9. Balancing eBooks and Physical Books Answers Mastering Chemistry Homework
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers Mastering Chemistry Homework
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers Mastering Chemistry Homework
 - Setting Reading Goals Answers Mastering Chemistry Homework
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers Mastering Chemistry Homework
 - Fact-Checking eBook Content of Answers Mastering Chemistry Homework

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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