

Analytical Chemistry MCQs: TYBSC SEM III

8. The solubility product K_{sp} of sparingly soluble Ag_2CrO_4 is 4×10^{-12} . The molar solubility of the salt is
- (a) 1.0×10^{-4} moles/lit
 - (b) 2.0×10^{-4} moles/lit
 - (c) 1.0×10^{-5} moles/lit
 - (d) 1.2×10^{-12} moles/lit
9. Solubility product of $AgCl$ is 1.44×10^{-4} moles/lit at 373K. The solubility of the $AgCl$ in boiling of water will be....
- (a) 0.72×10^{-4} moles/lit
 - (b) 0.72×10^{-2} moles/lit
 - (c) 1.20×10^{-2} moles/lit
 - (d) 1.20×10^{-4} moles/lit
10. Precipitation takes place when the ionic product....
- (a) Is zero
 - (b) Exceeds the solubility product
 - (c) Equal to solubility product
 - (d) Is less than the solubility product
11. The electromotive force is expressed in the unit of
- (a) Ohm
 - (b) Volt
 - (c) Ampere
 - (d) Coulomb
12. If the cell can provide electrical energy to an external system, it is called as
- (a) Galvanic cell
 - (b) Electrolytic cell
 - (c) Concentration cell
 - (d) Chemical cell
13. The current per unit area of electrode surface is known as....
- (a) Current efficiency
 - (b) Current yield
 - (c) Current density
 - (d) None of these

Analytical Chemistry Questions And Answers

Miguel Valcarcel



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Analytical Chemistry E. Hywel Evans, Mike E. Foulkes, 2019 Analytical Chemistry A Practical Approach is the only chemical analysis text with an emphasis on active learning giving students step by step guidance on how the key principles of analytical science are applied in a range of practical real world contexts

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Handbook of Food Analytical Chemistry, Volume 1 Ronald E. Wrolstad, Terry E. Acree, Eric A. Decker, Michael H. Penner, David S. Reid, Steven J. Schwartz, Charles F. Shoemaker, Denise M. Smith, Peter Sporns, 2005-09-16 Emphasizing effective state of the art methodology and written by recognized experts in the field the Handbook of Food Analytical Chemistry is an indispensable reference for food scientists and technologists to enable successful analysis Provides detailed reports on experimental procedures Includes sections on background theory and

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Charan,2017-08-01 The book now in its second edition provides a clear and concise understanding of the principles applications and limitations of the various techniques involved in analytical chemistry It motivates and prepares the students to face academic and research challenges in the field of analytical chemistry in performing analytical analysis and interpreting the results obtained The second edition while retaining the flow of chapters qualitative analysis quantitative analysis data analysis analysis of organic compounds separation and purification techniques electroanalytical techniques and spectroanalytical techniques introduces a new chapter on Thermoanalytical Techniques that discusses thermogravimetric analysis derivative thermogravimetric analysis and differential thermal analysis in detail Intended primarily as a text for the undergraduate and postgraduate students B Sc and M Sc of chemistry the book would also be of great benefit to the students who are appearing for NET and GATE examinations KEY FEATURES Provides clear introduction to all key analytical methods Uses a large number of illustrations to make each topic self explanatory Includes a large number of worked out problems for easy understanding of the concepts Contains numerous objective type questions short answer type questions and graded

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