

8. Buffer, passing question 10, is shown within the Earth's upper crust. Buffer is composed of the components of those of ascending velocity.

A sample of Buffer from a sample was analyzed to give the following composition of samples.

sample	composition (%)
1%	0.001
1%	0.01
1%	0.02

- (a) Define the term relative atomic mass.

The relative atomic mass of an element is the mass of one atom of the element divided by the mass of one atom of the element.

- (b) Calculate the relative atomic mass of the sample of buffer.

Give your answer to three decimal places.

$$(1 \times 0.001) + (1 \times 0.01) + (1 \times 0.02) = 0.031$$

0.031

Answer = 0.031 (3 d.p.)

- (c) Some studies, on early fossil, indicate that specimens were made up of long particles which could be used to determine the number of subatomic particles in atoms and in ions.

Complete the table to show the number of subatomic particles in the ^{12}C atom and $^{12}\text{C}^{2+}$ ion.

	protons	neutrons	electrons
^{12}C	6	6	6
$^{12}\text{C}^{2+}$	6	6	4



Answer To F321 Chemistry May Series

John Anderson



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