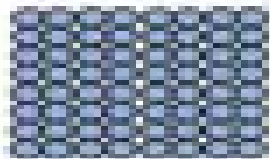


- 10) Choose one of the other substances from the menu on the right.
Experiment by adding and removing heat to the particles.
Draw and describe the substance in the table below.

Substance Chosen:

Neon

	Solid	Liquid	Gas
Diagram of the particles			
Description of particles (speed, movement, bouncing, etc.)	Slow, orderly, and close together. Very little movement. Only vibrating slightly.	Particles begin moving more and spreading out in the container. Higher energy.	Extremely high energy. Particles are spread out and bouncing all over the container.

- 10) Are particles in the solid state still moving? If so, how would you describe that movement?

Particles in the solid state are still moving, but only very slightly. The movement appears more like a vibration.

- 11) What happens to the behavior gas particles when more and more heat is added?

As more heat is added to particles in the gas state, their movement becomes more disorderly and energetic. They bounce even more quickly against the container.

- 12) What is the most notable difference between particles in the solid phase and the liquid phase?

Particles in the liquid phase contain more energy from heat. This causes the particles to spread out more and spin more quickly than particles in the solid state.

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