

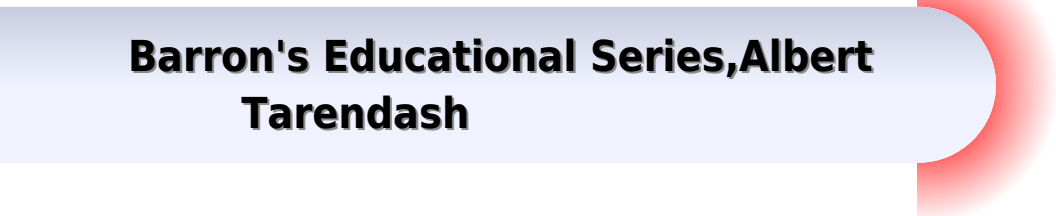
Chemical and Physical Properties

State whether each of the following is a physical property or a chemical property, and type of property.

Description	Physical/ Chemical	Type
1. The sulphur is yellow.	Physical	Colour
2. Iron reacts with sulphur to give heat and a flame.	Chemical	Combustibility; Reaction
3. Baking soda with acid produces bubbles of gas.	Chemical	Reaction with acid
4. Metal can be rolled into flat sheets.	Physical	Malleability
5. Oxygen is colourless, odourless and tasteless.	Physical	Colour, odour, taste
6. Nitrogen dioxide gas has a choking smell.	Physical	Odour
7. Helium is less dense than air, so a helium balloon floats.	Physical	Density
8. Wax candle burns in oxygen.	Chemical	Combustibility
9. An apple rots due to fungi.	Chemical	Reaction
10. Mercury metal is a liquid.	Physical	State
11. Iodine gas is purple.	Physical	Colour
12. Hydrofluoric acid is poisonous.	Chemical	Reaction
13. Gold is a shiny metal.	Physical	Lustre
14. Cesium is the only other gold-coloured metal.	Physical	Colour
15. Potassium burns with a purple flame to make a white powder.	Chemical	Combustibility
16. Copper nitrate is a blue crystal.	Physical	Colour, crystal form
17. Two colourless solutions mix to give a yellow solid precipitate.	Chemical	Reaction
18. Copper can be stretched into a thin wire.	Physical	Ductility
19. Copper metal reacts with nitric acid to make a brown gas.	Chemical	Reaction with acid
20. Salt, sodium chloride, is a white crystal that melts at 801 °C.	Physical	Colour, crystal form, melting point

Answers Physical And Chemical Properties

**Barron's Educational Series, Albert
Tarendash**



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Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day

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