

CH- ALDEHYDE, KETONES & CARBOXYLIC ACID

PRACTICE QUESTIONS

Q1. Convert Propene to Acetone.
Q2. Do the following conversions in not more than two steps: a) Ethyl Benzene to Benzoic Acid b) Propanone to Propene
Q3. Give reasons: a) Propanone is less reactive than ethanal towards nucleophilic addition reactions. b) $(\text{CH}_3)_2\text{CH-CHO}$ undergoes aldol condensation whereas $(\text{CH}_3)_3\text{C-CHO}$ does not.
Q4. Give a chemical test to distinguish between Butanal and Butan-2-one.
Q5. Convert Acetophenone to Benzoic Acid.
Q6. Give reasons: a) The α -hydrogen atoms of aldehydes and ketones are acidic in nature b) Propanone is less reactive than ethanal toward addition of HCN.
Q7. Convert: a) Toulene to Benzaldehyde b) Acetone to Propane
Q8. Give a chemical test to distinguish between Propanal and Propanone?
Q9. Arrange the following in increasing order of their reactivity towards nucleophilic addition reaction: a) $\text{C}_6\text{H}_5\text{COCH}_3$, $\text{CH}_3\text{-CHO}$, CH_3COCH_3 b) CH_3CHO , $\text{C}_6\text{H}_5\text{CHO}$, HCHO c) CH_3COCH_3 , $\text{C}_6\text{H}_5\text{COC}_6\text{H}_5$, CH_3CHO
Q10. Write the structure of the main products when propanal ($\text{CH}_3\text{-CH}_2\text{-CHO}$) reacts with the following reagents: a) Zn-Hg/conc. HCl b) $\text{H}_2\text{N-OH/H}^+$ c) HCN d) CH_3MgBr and then H_3O^+ e) $\text{H}_2\text{N-NHCONH}_2/\text{H}^+$ f) LiAlH_4
Q11. Write the products formed when ethanal reacts with the following reagents: a) CH_3MgBr and then H_3O^+ b) Zn-Hg/conc. HCl c) $\text{C}_6\text{H}_5\text{CHO}$ in the presence of dilute NaOH
Q12. Out of $\text{CH}_3\text{CH}_2\text{-CO-CH}_2\text{-CH}_3$ and $\text{CH}_3\text{CH}_2\text{-CH}_2\text{-CO-CH}_3$, which gives iodoform test?

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