Name:

Chemistry Practice: Balancing Chemical Equations

 OO_2

- Balance the equations below. Remember this means only using coeeficients. All formulas are written correctly.
 - a.) 2 NaHCO₃ Na₂CO₃ + H₂O +
 - b.) Zn + S ZnS
 - c.) 2H₂ + O₂ → ZH₂O
 - d.) 2 FeO + C ZFe + CO₂
 - e.) $Cl_2 + ZKI \longrightarrow ZKCI + I_2$
 - f.) $B_2O_3 + 3Mg 3MgO + ZB$

 - (h) 5H₂O + Na₂B₄O₇ + H₂SO₄ → 4/H₃BO₃ + Na₂SO₄
 - i.) $CaC_2 + 2H_2O$ \longrightarrow $Ca(OH)_2 + C_2H_2$
 - j.) NH₄NO₃ N₂O +2H₂O
 - k.) $H_2S + O_2$ $SO_2 + H_2O$
 - H₂SO₄ + 2NaCl → ZHCl + Na₂SO₄

 - n.) 3 BaO + Z Al _____ 3 Ba + Al₂O₃
 - o.) 3 O₂ -> 2 O₃
 - p.) $SiO_2 + ZC$ \longrightarrow ZCO + Si
 - q.) 2 PbS + 3 O₂ ____ 2 PbO + 2 SO₂
 - r.) 2 HgO → 2 Hg + O₂

 - 1.) $C_3H_8 + 5O_2 \rightarrow 3CO_2 + 4H_2O$

 - v.) P₂O₅ + 6 NaOH 2 Na₃PO₄ + 3 H₂O
 - w.) $3 \text{ BaCl}_2 + \text{Al}_2(SO_4)_3 \longrightarrow 3 \text{ BaSO}_4 + 2 \text{AlCl}_3$
 - x.) 4/NH₃ + 5 O₂ 6 H₂O + 4/NO
 - y.) Z Na₂O₂ + 2 H₂O → 4/ NaOH + 4/ O₂
 - z.) $Sn(ClO_3)_4$ \longrightarrow $SnCl_4 + 6 O_2$

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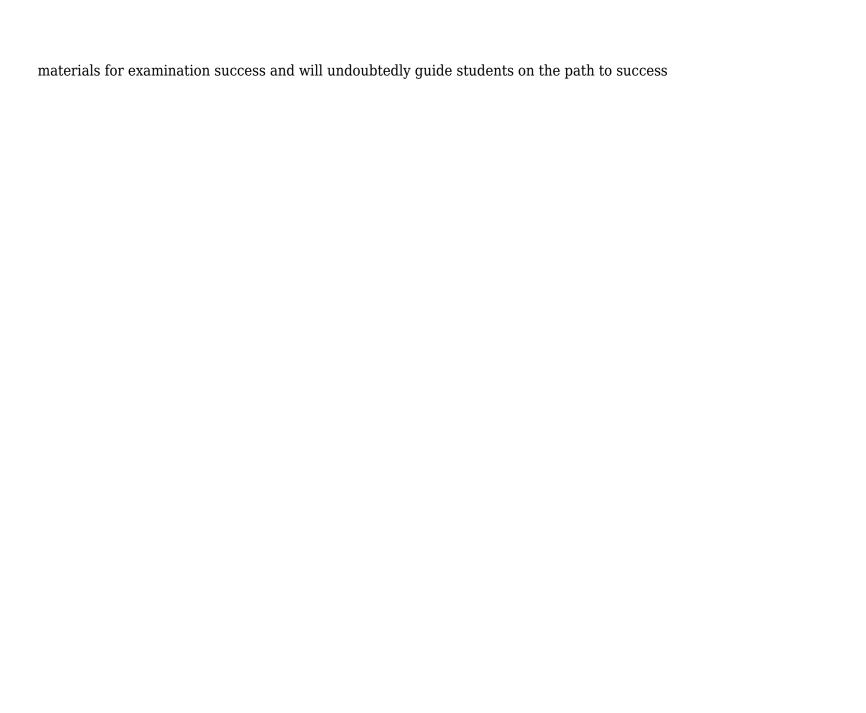
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