Name:

Chemistry Practice: Balancing Chemical Equations

 OO_2

- Balance the equations below. Remember this means only using coeeficients. All formulas are written correctly.
 - a.) 2 NaHCO₃ Na₂CO₃ + H₂O +
 - b.) Zn + S \longrightarrow ZnS
 - c.) 2H₂ + O₂ ZH₂O
 - d.) 2 FeO + C 2Fe + CO₂
 - e.) Cl₂ + ZKI

 ▶ 2_KCl + I₂
 - f.) $B_2O_3 + 3Mg \longrightarrow 3MgO + ZB$
 - g.) Ca +2 H₂O

 → Ca(OH)₂ + 2 H₂
 - (h) 5H₂O + Na₂B₄O₇ + H₂SO₄ → 4/H₃BO₃ + Na₂SO₄
 - i.) $CaC_2 + 2H_2O$ \longrightarrow $Ca(OH)_2 + C_2H_2$
 - j.) NH₄NO₃ N₂O +2H₂O
 - k.) $H_2S + O_2$ $SO_2 + H_2O$
 - H₂SO₄ + 2NaCl → ZHCl + Na₂SO₄

 - n.) 3 BaO + Z Al _____ 3 Ba + Al₂O₃
 - o.) 3 O₂ → 2 O₃
 - p.) $SiO_2 + ZC$ \longrightarrow 2 CO + Si
 - q.) 2 PbS + 3 O₂ ____ 2 PbO + 2 SO₂
 - r.) 2 HgO → 2 Hg + O₂
 - s.) $C_{12}H_{22}O_{11} + /ZO_2 /Z_CO_2 + //H_2O$
 - t) $C_3H_8 + 5O_2 \rightarrow 3CO_2 + 4H_2O$

 - v.) P₂O₅ + 6 NaOH 2 Na₃PO₄ + 3 H₂O
 - w.) 3 BaCl₂ + Al₂(SO₄)₃ 3 BaSO₄ + 2 AlCl₃
 - x.) 4/NH₃ + 5 O₂ 6 H₂O + 4/NO
 - y.) Z Na₂O₂ + 2 H₂O → 4/NaOH + 4/O₂
 - z.) $Sn(ClO_3)_4$ \longrightarrow $SnCl_4 + 6 O_2$

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