

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

1. Explain how a chemical equation is written.
2. Explain how to write a balanced equation.
3. Write the steps for writing a balanced chemical equation.

Vocabulary

- | | |
|----------------------|----------------------|
| 1. chemical equation | 4. coefficient |
| 2. stoichiometric | 5. balanced equation |
| 3. catalyst | |

Part A Completion

Use the vocabulary words in **Part A** to complete the following sentences. Write the word or phrase in the space provided.

1. A chemical equation does include information about the _____ of the chemical substances that react and the _____ of the substances that are produced.
2. The law of conservation of mass states that the _____ of matter is conserved in a chemical reaction.
3. A chemical equation is balanced when the _____ of each element is the same on both sides of the equation.
4. A chemical equation is balanced when the _____ of each element is the same on both sides of the equation.
5. A chemical equation is balanced when the _____ of each element is the same on both sides of the equation.
6. A chemical equation is balanced when the _____ of each element is the same on both sides of the equation.
7. A chemical equation is balanced when the _____ of each element is the same on both sides of the equation.
8. A chemical equation is balanced when the _____ of each element is the same on both sides of the equation.
9. A chemical equation is balanced when the _____ of each element is the same on both sides of the equation.
10. A chemical equation is balanced when the _____ of each element is the same on both sides of the equation.

1. coefficient
2. stoichiometric
3. catalyst
4. balanced equation
5. balanced equation
6. balanced equation
7. balanced equation
8. balanced equation
9. balanced equation
10. balanced equation

11 1 Section Review Describing Chemical Reactions

Answer Key

Lei Shi



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Unveiling the Energy of Verbal Beauty: An Psychological Sojourn through **11 1 Section Review Describing Chemical Reactions Answer Key**

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