

11

CHEMICAL REACTIONS

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329)

This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.

► Writing Chemical Equations (pages 321–323)

1. A chemical reaction occurs when one or more _____ change into one or more new substances called _____.
2. The arrow in a reaction means _____.
3. Is the following sentence true or false? When there are two or more reactants or products, they are separated by an arrow. _____
4. Write a word equation that describes the following reactions.
 - a. Acetylene reacts with oxygen to produce carbon dioxide and water.

 - b. When heated, mercury(II) oxide reacts to form mercury and oxygen.

5. What is a chemical equation?

6. A chemical reaction that shows only the formulas, but not the relative amounts of the reactants and products is a(n) _____.
7. Identify the reactant(s) and product(s) in the chemical equation $\text{Li} + \text{Br}_2 \longrightarrow \text{LiBr}$.
 - a. reactant(s) _____
 - b. product(s) _____
8. Circle the letter of each statement that is true about a catalyst.
 - a. A catalyst is the new material produced as a result of a chemical reaction.
 - b. A catalyst is not used up in a chemical reaction.
 - c. A catalyst adds heat to a chemical reaction.
 - d. A catalyst speeds up a chemical reaction.

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Daniel F McAuley



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Focus on Physical Science California Edition Michael J. Padilla, 2007 Elements of Chemical Reaction Engineering H. Scott Fogler, 2020-08-18 The Definitive Guide to Chemical Reaction Engineering Problem Solving With Updated Content and More Active Learning For decades H Scott Fogler's Elements of Chemical Reaction Engineering has been the world's dominant chemical reaction engineering text This Sixth Edition and integrated Web site deliver a more compelling active learning experience than ever before Using sliders and interactive examples in Wolfram Python POLYMATH and MATLAB students can explore reactions and reactors by running realistic simulation experiments Writing for today's students Fogler provides instant access to information avoids extraneous details and presents novel problems linking theory to practice Faculty can flexibly define their courses drawing on updated chapters problems and extensive Professional Reference Shelf web content at diverse levels of difficulty The book thoroughly prepares undergraduates to apply chemical reaction kinetics and physics to the design of chemical reactors And four advanced chapters address graduate level topics including effectiveness factors To support the field's growing emphasis on chemical reactor safety each chapter now ends with a practical safety lesson Updates throughout the book reflect current theory and practice and emphasize safety New discussions of molecular simulations and stochastic modeling Increased emphasis on alternative energy sources such as solar and biofuels Thorough reworking of three chapters on heat effects Full chapters on nonideal reactors diffusion limitations and residence time distribution About the Companion Web Site umich.edu/elements/6e/index.html Complete PowerPoint slides for lecture notes for chemical reaction engineering classes Links to additional software including POLYMATH™ MATLAB™ Wolfram Mathematica™ AspenTech™ and COMSOL™ Interactive learning resources linked to each chapter including Learning Objectives Summary Notes Web Modules Interactive Computer Games Solved Problems FAQs additional homework problems and links to Learncheme Living Example Problems unique to this book that provide more than 80 interactive simulations allowing students to explore the examples and ask what if questions Professional Reference Shelf which includes advanced content on reactors weighted least squares experimental planning laboratory reactors pharmacokinetics wire gauze reactors trickle bed reactors fluidized bed reactors CVD boat reactors detailed explanations of key derivations and more Problem solving strategies and insights on creative and critical thinking Register your book for convenient access to downloads updates and or corrections as they become available See inside book for details *Density Functional Theory* Daniel Glossman-Mitnik, 2019-01-30 Density Functional Theory or DFT for short is a potent methodology useful for calculating and understanding the molecular and electronic structure of atoms molecules clusters and solids Its use relies not only in the ability to calculate the molecular properties of the species of interest but also provides interesting concepts that allow a better comprehension of the chemical reactivity of the studied systems This book represents an attempt to present examples on the utility of DFT for the understanding of the chemical reactivity through descriptors that constitute

the basis of the so called Conceptual DFT sometimes also named as Chemical Reactivity Theory as well as the application of the theory and its related computational procedures in the determination of the molecular properties of different systems of academic and industrial interest *Chemical Reactivity Theory* Pratim Kumar Chattaraj, 2009-02-23 In the 1970s Density Functional Theory DFT was borrowed from physics and adapted to chemistry by a handful of visionaries Now chemical DFT is a diverse and rapidly growing field its progress fueled by numerous developing practical descriptors that make DFT as useful as it is vast With 34 chapters written by 65 eminent scientists from 13 different countries **A TEXTBOOK OF PHARMACEUTICAL**

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