

CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

*Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.*

This paper is divided into four sections – A, B, C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B,

Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets [].

Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

$$\begin{aligned}\text{Gas constant } R &= 1.987 \text{ cal deg}^{-1} \text{ mol}^{-1} = 8.314 \text{ JK}^{-1} \text{ mol}^{-1} \\ &= 0.0821 \text{ dm}^3 \text{ atm K}^{-1} \text{ mol}^{-1}\end{aligned}$$

$$1 \text{ l atm} = 1 \text{ dm}^3 \text{ atm} = 101.3 \text{ J} \quad 1 \text{ Faraday} = 96500 \text{ coulombs.}$$

$$\text{Avogadro's number} = 6.023 \times 10^{23}$$

SECTION A – 14 MARKS

Question 1

- (A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1]
brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid, high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into _____.
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and oxidation state of the complex $K_4[Fe(CN)_6]$ are _____ and _____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives _____ and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

12 Question Paper For Chemistry

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12 Question Paper For Chemistry:

12th Standard Chemistry English Medium Model Questions Papers - Tamil Nadu State Board Syllabus Mukil E Publishing And Solutions Pvt Ltd,2021-06-05 12th Standard Chemistry English Medium Tamil Nadu State Board solutions guide For the first time in Tamil Nadu Technical books are available as ebooks Students and Teachers make use of it *CBSE Class 12th Chemistry (10 Most Likely Question Papers with Solution) By Career Point Kota* Career Point Kota,2020-07-28 The CBSE has made certain changes in the assessment structure from the session 2019 20 onwards In the new scheme of examination CCE and term system has been replaced with the Internal Assessment Single Annual Exam by CBSE itself Single exam conducted by CBSE will carry 80 marks whereas 20 marks are left to the schools for internal assessment CBSE has issued detailed guidelines on how the internal marks will be divided among different activities From 2019 onwards there will be internal choices in board examinations with increased internal options in the question paper Considering this change now a student has to prepare accordingly for board examinations The new assessment format brought with it excitement as well as anxiety And to help them prepare and excel in their CBSE board examination Career Point Kota has developed a series of 10 Most Likely Question Papers with Solutions The Key Features of Most Likely Question Papers with Solutions Series New OBJECTIVE TYPE question in each paper Syllabus of CBSE 2019 20 Based on the latest CBSE Syllabus Pattern Mind map of each chapter is given to visualize and help acquire a better understanding Important terms facts formulae and quick revision tips are given Covers questions asked in previous year board exams Toppers Answer Sheet as released by CBSE to understand the scoring technique We hope this book will gratify students need for the new CBSE pattern board exam and smoothen their path to success We wish to utilize the opportunity to place on record our special thanks to all the members of the Content Development team for their efforts to create this wonderful book Last 5+1 Year's CBSE Class 12th Chemistry Solved Question Papers - eBook JagranJosh,2017-11-21 This Combo Package prepared by CBSE Exam experts at Jagranjosh com is a kind of must have for the students appearing for Class12th Chemistry Paper in the coming CBSE Board 2018 Exam 1 This Combo Package includes CBSE Class 12 Chemistry Solved Question Paper 2017 CBSE Class 12 Chemistry Solved Question Paper 2016 CBSE Class 12 Chemistry Solved Question Paper 2015 CBSE Class 12 Chemistry Solved Question Paper 2014 CBSE Class 12 Chemistry Solved Question Paper 2013 CBSE Class 12 Chemistry Solved Question Paper 2012 2 The Package strictly follows the pattern of CBSE Class 12th Syllabus 4 It also contains the detailed explanation for each question solved 5 It will help you strengthen the concepts at class 12th level 6 This Package will surely Build your confidence to score excellent marks in following Board Exam Paper Key Feature Free Class 12th Chemistry 2012 Solved Paper ebook Ideal to understand the exam pattern Will give a clear idea of how to study and what to study for the exam *Oswaal CBSE Sample Question Papers Class 12 Chemistry (For 2024 Exam)* Oswaal Editorial Board,2023-08-03 Description of the product Fresh Relevant with 2024 CBSE SQP Fully Solved Analyzed Score Boosting Insights with 500 Questions 1000 Concepts Insider Tips

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