

* Solid state *

- * **Solid** → have definite mass, volume, shape
 → short intermolecular distance
 → strong intermolecular force
 → incompressible & rigid
 → particle have fixed position & only oscillate about their position.

* solid have 2 types

crystalline

→ The solid which have definite order of arrangement of particle
 ex. NaCl, Quartz

Amorphous

→ The solid have irregular arrangement of particle
 ex. glass, rubber, plastic, Quartz glass

properties	crystalline	Amorphous
shape	definite geometric shape	irregular shape.
Melting	Melt at sharp temp.	Gradually soften over range of temp.
cleavage	when cut newly generated surface are plain & smooth	when cut new generated surface are irregular.
heat of fusion	definite heat of fusion	not have definite heat of fusion.
Anisotropy	Anisotropic (properties change in differ. direction)	isotropic (properties same in all direction)
Nature	True solid	Pseudo solid or super cooled liquid
order of arrangement	Long range order	only short range order.

12th Science Chemistry Target Solid State Notes

Minjie Lin



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Table of Contents 12th Science Chemistry Target Solid State Notes

1. Understanding the eBook 12th Science Chemistry Target Solid State Notes
 - The Rise of Digital Reading 12th Science Chemistry Target Solid State Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying 12th Science Chemistry Target Solid State Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 12th Science Chemistry Target Solid State Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from 12th Science Chemistry Target Solid State Notes
 - Personalized Recommendations
 - 12th Science Chemistry Target Solid State Notes User Reviews and Ratings
 - 12th Science Chemistry Target Solid State Notes and Bestseller Lists
5. Accessing 12th Science Chemistry Target Solid State Notes Free and Paid eBooks
 - 12th Science Chemistry Target Solid State Notes Public Domain eBooks
 - 12th Science Chemistry Target Solid State Notes eBook Subscription Services
 - 12th Science Chemistry Target Solid State Notes Budget-Friendly Options
6. Navigating 12th Science Chemistry Target Solid State Notes eBook Formats

- ePub, PDF, MOBI, and More
 - 12th Science Chemistry Target Solid State Notes Compatibility with Devices
 - 12th Science Chemistry Target Solid State Notes Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 12th Science Chemistry Target Solid State Notes
 - Highlighting and Note-Taking 12th Science Chemistry Target Solid State Notes
 - Interactive Elements 12th Science Chemistry Target Solid State Notes
 8. Staying Engaged with 12th Science Chemistry Target Solid State Notes
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 12th Science Chemistry Target Solid State Notes
 9. Balancing eBooks and Physical Books 12th Science Chemistry Target Solid State Notes
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 12th Science Chemistry Target Solid State Notes
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine 12th Science Chemistry Target Solid State Notes
 - Setting Reading Goals 12th Science Chemistry Target Solid State Notes
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of 12th Science Chemistry Target Solid State Notes
 - Fact-Checking eBook Content of 12th Science Chemistry Target Solid State Notes
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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