

# Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON-METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.)  $H_2$  hydrogen is diatomic



Single Bond

2.)  $F_2$  fluorine is diatomic



3.)  $O_2$  oxygen is diatomic



Double Bond

4.)  $N_2$  nitrogen is diatomic. Is this a triple bond?



5.)  $BF_3$  you need 3 fluorine atoms here



6.) Ammonia  $NH_3$  hint: how many hydrogen atoms are needed?



7.) Carbon dioxide  $CO_2$



8.) Methane  $CH_4$  careful here 4 hydrogen atoms needed



9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!



10.)  $SO_2$  hint: one pair of electrons from sulfur must be slit up for this one to work.



# Answer Key To The Nature Of Covalent Bonding

**J Dewey**



## **Answer Key To The Nature Of Covalent Bonding:**

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