

SCH4U Unit 4 - Quiz – Equilibrium

Player: _____ Period: _____

120 130 140 150

160 170

180

190 200

210

220 230 240 250

$$K_{eq} = \frac{[products]}{[reactants]} \quad K_{eq} = \frac{[C]^c [D]^d}{[A]^a [B]^b}$$

$$K_{eq} = \frac{P_{CO_2}^2 P_{H_2O}}{P_{CH_4}^2 P_{O_2}} \quad K_{eq} = \frac{P_{CO_2}^2 P_{H_2O}}{(P_{CH_4})^2 (P_{O_2})^2}$$

gpt (g/mol) x vol (L) gpt (L) = mol (gpt) (L) gpt (L) = mol (gpt) (L) x 10³ L x 10³ L

Question 1:

1. Hydrogen sulfide, H₂S, is a slightly poisonous gas. At 100°C, an equilibrium mixture contains 0.020 mol of hydrogen, H₂, 0.010 mol of hydrogen sulfide, and an undetermined amount of sulfur, S₂. The reaction is:



If the volume of the mixture is 1.0 L and 0.020 mol of H₂ is present at equilibrium:

- What is the concentration of S₂ at equilibrium?
- What is the concentration of H₂ at equilibrium?
- What is the concentration of H₂ at equilibrium?
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- What is the concentration of H₂ at equilibrium?
- What is the concentration of H₂ at equilibrium?
- What is the concentration of H₂ at equilibrium?

	2H ₂ (g)	S ₂ (g)	2H ₂ S(g)
Initial (mol/L)			
Change (mol/L)			
Equilibrium (mol/L)	0.020	0.010	

What is the concentration of H₂ at equilibrium?

Answers For Sch4u Unit 4

Michael Hoelscher



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