

| Question Number | Acceptable Answers                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                         | Reject  | Mark      |       |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |
|-----------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|---------|-----------|-------|-----|-------|---|-------|------|--------|-------|---|------|-----|--------|-------|---|-------|------|----------|-------|---|-------|------|-----------|-------|--|---|
| 23(a)           | <table><tr><th>Element</th><th>%</th><th></th><th>mol</th><th>Ratio</th></tr><tr><td>C</td><td>40.44</td><td>÷ 12</td><td>= 3.37</td><td>2.999</td></tr><tr><td>H</td><td>7.87</td><td>÷ 1</td><td>= 7.87</td><td>7.004</td></tr><tr><td>O</td><td>35.96</td><td>÷ 16</td><td>= 2.2475</td><td>2.000</td></tr><tr><td>N</td><td>15.73</td><td>÷ 14</td><td>= 1.12357</td><td>1.000</td></tr></table> <p style="text-align: center;">(1)                      (1)</p> <p>Empirical formula = C<sub>3</sub>H<sub>7</sub>O<sub>2</sub>N (1)<br/>Symbols in any order<br/>Stand alone mark<br/>No TE on incorrect ratio<br/>I IGNORE significant figure and rounding errors except 1sf in mole calculation</p> | Element | %         |       | mol | Ratio | C | 40.44 | ÷ 12 | = 3.37 | 2.999 | H | 7.87 | ÷ 1 | = 7.87 | 7.004 | O | 35.96 | ÷ 16 | = 2.2475 | 2.000 | N | 15.73 | ÷ 14 | = 1.12357 | 1.000 |  | 3 |
| Element         | %                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                          |         | mol       | Ratio |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |
| C               | 40.44                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | ÷ 12    | = 3.37    | 2.999 |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |
| H               | 7.87                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                       | ÷ 1     | = 7.87    | 7.004 |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |
| O               | 35.96                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | ÷ 16    | = 2.2475  | 2.000 |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |
| N               | 15.73                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | ÷ 14    | = 1.12357 | 1.000 |     |       |   |       |      |        |       |   |      |     |        |       |   |       |      |          |       |   |       |      |           |       |  |   |

| Question Number | Acceptable Answers                                                                                                       | Reject | Mark     |
|-----------------|--------------------------------------------------------------------------------------------------------------------------|--------|----------|
| <b>23(b)(i)</b> | Peak at $m/e = 89$ labelled $\text{M}^+$<br><br>ALLOW<br>Any clear label e.g. $\text{C}_3\text{H}_7\text{O}_2\text{N}^+$ |        | <b>1</b> |

| Question Number  | Acceptable Answers                                                                                                                                                                                | Reject                                                                                 | Mark     |
|------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------|----------|
| <b>23(b)(ii)</b> | $M_r = 89 = M_r(\text{C}_3\text{H}_7\text{O}_2\text{N})$ so molecular formula is $\text{C}_3\text{H}_7\text{O}_2\text{N}$<br>Symbols in any order<br><br>IGNORE structural and displayed formulae | Answer with no explanation<br>$M_r = 90$<br>$\text{C}_3\text{H}_7\text{O}_2\text{N}^+$ | <b>1</b> |

# Answer Sheet Of 2014 C12 Ial Paper

**D Keegan**



**Answer Sheet Of 2014 C12 Ial Paper:**

## **Answer Sheet Of 2014 C12 Ial Paper** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Answer Sheet Of 2014 C12 Ial Paper**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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