

## CHAPTER 7 REVIEW

***Chemical Formulas and Chemical Compounds*****SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. \_\_\_\_\_ In a Stock system name such as iron(III) sulfate, the Roman numeral tells us
  - (a) how many atoms of Fe are in one formula unit.
  - (b) how many sulfate ions can be attached to the iron atom.
  - (c) the charge on each Fe ion.
  - (d) the total positive charge of the formula unit.
2. \_\_\_\_\_ Changing a subscript in a correctly written chemical formula
  - (a) changes the number of moles represented by the formula.
  - (b) changes the charges on the other ions in the compound.
  - (c) changes the formula so that it no longer represents the compound it previously represented.
  - (d) has no effect on the formula.
3. The explosive TNT has the molecular formula  $C_7H_5(NO_2)_3$ .
  - \_\_\_\_\_ a. How many elements make up this compound?
  - \_\_\_\_\_ b. How many oxygen atoms are present in one molecule of  $C_7H_5(NO_2)_3$ ?
  - \_\_\_\_\_ c. How many atoms in total are present in one molecule of  $C_7H_5(NO_2)_3$ ?
  - \_\_\_\_\_ d. How many atoms are present in a sample of  $2.0 \times 10^{23}$  molecules of  $C_7H_5(NO_2)_3$ ?
4. How many atoms are present in each of these formula units?
  - \_\_\_\_\_ a.  $Ca(HCO_3)_2$
  - \_\_\_\_\_ b.  $C_{12}H_{22}O_{11}$
  - \_\_\_\_\_ c.  $Fe(ClO_2)_3$
  - \_\_\_\_\_ d.  $Fe(ClO_3)_2$
5. \_\_\_\_\_
  - \_\_\_\_\_ a. What is the formula for the compound dinitrogen pentoxide?
  - \_\_\_\_\_ b. What is the Stock system name for the compound  $FeO$ ?
  - \_\_\_\_\_ c. What is the formula for sulfurous acid?
  - \_\_\_\_\_ d. What is the name for the acid  $H_3PO_4$ ?

# Answers Review Chemical Formulas And Chemical Compounds

**G Orfield**



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